

SYLLABUS

CHEMISTRY 101

SPRING 2017

LECTURES: MWF 9:10AM & 12:10PM

Fulmer 226

INSTRUCTORS: Dr. ChulHee Kang Fulmer 264 335-1409 chkang@wsu.edu

Office Hours: 1:00PM-2:00PM MWF, or by appointment.

Dr. Krista Nishida Fulmer 317A 335-4935 krista_nishida@wsu.edu

Office Hours: 11:00AM-12:00PM MWF, or by appointment.

COURSE WEB SITE: Blackboard Learn: [https:// learn.wsu.edu/](https://learn.wsu.edu/).

GRADING:	3 "midterm" exams	300	GRADE RANGES: (minimum points to achieve)		
	10 tutorial quizzes (best 8)	120	900 points	A	730 points C+
	Homework problem sets	70	870 points	A-	700 points C
	Reading Assignments	70	830 points	B+	670 points C-
	In Lecture Assignments	70	800 points	B	630 points D+
	Laboratory reports & worksheets	170	770 points	B-	600 points D
	Final Exam	<u>200</u>			Less than 600 points: F
	TOTAL	1000			

MIDTERM EXAMS: Thursday Feb 9 6:00– 7:00 pm (Ch 1-4 + Lab Wkst #2 & Expt #1)
Thursday Mar 9 6:00– 7:00 pm (Ch 1 & 5-7.3 + Expts #2, 4, 10, Wkst #1)
Thursday Apr 13 6:00– 7:00 pm (Ch 7.4-10.6 + Experiments 6, 7, & 8)

FINAL EXAM Monday May 1 7:00pm–10:00 pm (Comprehensive + all Expts/worksheets)

TEXT General, Organic and Biological Chemistry; McMurry, Ballantine, Hoeger & Peterson; 8th edition (2017). Pearson/Prentice Hall. This is a custom print of Chapters 1-13. Access for the Mastering Chemistry homework system is bundled with new copies of this. Both the text and Mastering Chemistry access are required for this course. Individual access codes for the Mastering Chemistry system can be purchased via the MasteringChemistry link on the course Blackboard website (see Homework).

LAB TEXT: *Chemistry 101 Laboratory Manual* by WSU Chemistry Department, Star Publishing (2016, required).

LABORATORY NOTEBOOK: Duplicating with numbered pages. (Sold in Fulmer 318 the 1st and 2nd week of class.)

GOGGLES: Required by State Law. (Sold in Fulmer 318 the 1st and 2nd week of class.)

LABORATORY COAT: Optional but recommended. A strict dress code is enforced in the laboratories. NO SHORTS, NO SHORT SKIRTS, NO SANDALS, NO BARE MIDRIFTS. (See laboratory dress code.)

CALCULATORS: You are expected to have and to be able to use a scientific calculator. Graphing calculators are allowed but not required. The use of any stored information/programs in a programmable calculator will be considered cheating. Calculators with a full QWERTY keyboard (such as the TI-92 or Voyage 200); PDAs; palmtops, laptops, handheld computers, and cell phone/calculator combinations may not be used during quizzes and examinations. You are responsible for bringing your calculator to all tutorials, lectures, labs and exams.

ONLINE COMPONENTS: We will be using Blackboard Learn for course management and online information. This can be accessed via <https://learn.wsu.edu>. Use your WSU network ID and password to log in. All online aspects of the course including homework sets (MasteringChemistry), reading assignments (MasteringChemistry), and In Lecture Assignments (Learning Catalytics) should be accessed via the links on the Blackboard site. All e-mail communications to the course instructor and TAs should be via the Blackboard mail tool. Confidential information such as scores and grades may not be transmitted via unsecured email.

READING ASSIGNMENTS: There will be reading assignments due before each lecture starting with the Wednesday, January 11th lecture (lecture #1). These consist of assigned sections of the text and questions intended to evaluate your understanding of this material. The reading assignments are available on the MasteringChemistry site accessed via the Blackboard course site. Each assignment must be completed by 2:00AM on the day of the associated lecture. Each reading assignment will be available a week before they are due (except reading assignments 1 and 2). Each reading assignment is worth 3 points. Your score will be determined by multiplying your percent correct by 3 and rounding to the nearest ½ point. A maximum of 70 of the 123 points from the reading assignments will count toward the course grade. A portion of any reading assignment points obtained in excess of 70 points will be counted as extra credit at the end of the

semester. It is important to note that the completion of these assignments is independent of lecture attendance. If you are sick or out of town, it is still possible to complete the assignments.

LECTURES: Lectures must be attended on a regular basis. You will be expected to read the textbook AHEAD of coming to class. Lectures will supplement and clarify the information from your text rather than reiterate it. Lectures will focus on problem solving and demonstrations of chemical reactions. There will be in-lecture problem assignments via the LearningCatalytics website (accessible via the course Blackboard site). These will be graded and counted towards your total point score. These in-lecture assignments will be unannounced, cannot be made up, and constitute the only possible 'extra credit' points in this course. *Bring a calculator and web-enabled device to all lectures.* You are encouraged to form collaborative study groups and to sit with your group members during lecture.

IN-LECTURE ASSIGNMENTS: Most lectures will include problem assignments via the LearningCatalytics website (accessible via the course Blackboard site). These assignments will be identified as "In-Lecture Assignments" in the grade book. These sessions are interactive and will require a Wi-Fi enabled device such as a smart-phone, laptop, or tablet. Each in-lecture assignment will be worth 4 course points and will be graded on both participation (25%) and correctness of answers (75%). There will be in excess of 150 points available from in-lecture assignments. A maximum of 70 points from the in-lecture assignments will count toward the course grade. A portion of any in-lecture assignment points obtained in excess of 70 points will be counted as extra credit at the end of the semester. Each in-lecture assignment will be available only during a portion of the associated lecture. There will be no make-up opportunities for in-lecture assignments. If you are not present for lecture for any reason, you will not have the opportunity to get those 4 points. If your Wi-Fi device fails during the assignment, you will not have an opportunity to get the points for the portion of the assignment you are not able to complete. Make certain that the batteries are fully charged before arriving for lecture.

HOMEWORK: There will be weekly homework assignments consisting of a minimum of ten questions each. These assignments are administered by Mastering Chemistry and accessed through the course Blackboard site. You should have an access code for Mastering Chemistry bundled with your text. Please follow the attached instructions and use the access code to register for the system. If you purchased a text that did not include an access code, follow the attached instructions to purchase a code and register for the system.

A new homework assignment will be made available each week (no later than 5:00PM each Monday). Each assignment must be completed by 8:00PM the following Monday in order to obtain full credit. Partial credit may be obtained up to 5 days after the due time (available points decrease by 20% each day). The due date/time for each assignment will be listed with the assignment on the homework site. Each homework set will be pro-rated to have a value of 8 course points. Thus there will be 120 homework points available this semester. A maximum of 70 points from homework will count toward the final grade. A portion of any homework points obtained in excess of 70 points will be counted as extra credit at the end of the semester.

TUTORIALS: These are small classroom meetings associated with your laboratory section and led by your TA. Students who miss tutorial will **not** be allowed into the lab. Quizzes are given in tutorial most weeks (see the course schedule). Tutorials are interactive problem solving sessions driven by your questions. Bring your text, lab manual and calculator to tutorial. Pre-labs and lab reports are due at the start of tutorial. Help with pre-labs and lab reports will not be available in tutorial as they must be completed before attending tutorial. **Tutorial sessions are never canceled!** If your TA fails to arrive for a tutorial section, send one person to contact Ryan Rice (Fulmer 309) or Dr. Finnegan immediately. All others must remain in the tutorial room until the TA or a substitute arrives. Students who leave tutorial under these circumstances will forfeit all points associated with that tutorial/laboratory session (lab report, lab, and quiz).

QUIZZES: There will be ten 15-point quizzes of which the best eight will count toward the final grade. Quizzes are given in tutorial. Quizzes will cover lecture, homework and laboratory material. You will be allowed to prepare a single 3" x 5" card containing your HAND-WRITTEN notes for use during each of the quizzes. No other handwritten material and no printed or photocopied material may be used during the quiz, except for an approved periodic table (the table that appears on the back of your lab manual). Quiz #10 will be given during closed week and qualifies as a make-up quiz under academic regulation #79.

EXAMS: There will be three midterm exams and a comprehensive final. All exams will be multiple-choice. You will be responsible for bringing a calculator and a pencil to all exams. A bubble-in answer sheet will be provided. No notes or books are allowed. Exams may be given in rooms other than the regular classroom. These rooms will be announced. No make-up exams will be given. If you are unable to take a scheduled midterm exam for academic reasons beyond your control, you will be allowed to schedule the exam at an earlier time. Midterm exams missed due to illness will be

excused, with the other exams plus the final pro-rated to count for more. **Evening exams take precedence over all other university activities.**

LABORATORIES: Your course section includes a lecture time and a laboratory time. This is a laboratory GER course, thus the laboratory must be completed in order to pass the course. Thus, obtaining a score of zero for 3 or more experiments will result in an F for the course.

Make-up labs: Labs missed for reasons beyond your control, may be made up, on a space available basis, in the same week that the lab is missed. You will be allowed to make up a maximum of two labs per semester in this manner. Permission for a make-up lab must be obtained, in writing, from the Chemistry Office, Fulmer 319A. The permission slip will be collected and signed by the make-up TA. **We cannot guarantee that make-up space will be available.** If you know in advance that you will miss a lab, visit Fulmer 319A as soon as possible in order to maximize the chance that make-up space will be available. **If make-up space is not available:** a report may be written from data supplied by the instructor. Such reports are due at the normal time (in tutorial one week after you should have attended lab) and will be worth no more than ½ credit.

Pre-laboratory assignments: Pre-laboratory assignments are due at the start of the tutorial. Students who fail to submit a complete pre-lab assignment at this time will be assessed a late penalty on the full report and be required to complete the pre-lab assignment before they are admitted to lab. The student will not be given extra time in the laboratory to make up for laboratory time spent completing the pre-lab. Students who attend tutorial without a complete pre-lab assignment will not be allowed to arrange a make-up lab for that experiment.

Laboratory procedure: Students are to perform the experiments individually unless the laboratory manual specifically requires partners for the experiment being performed. Each student is expected to record all data and observations for each experiment directly into their own laboratory notebook. Data may not be recorded on loose, 'scratch' paper then transferred to the notebook. Submission of identical data by two or more students who are not assigned to be laboratory partners will be considered cheating. Appropriate penalties will be applied to all parties. You are required to get your TA's signature on your data and observations and submit the original data/observation pages before you leave lab. Failure to do so will result in zero credit for that experiment.

Laboratory dress code: For your safety, a strict dress code will be enforced in the laboratory. Failure to comply with the dress code will result in expulsion from the laboratory and a consequent score of zero for that experiment. The dress code requires that you be fully clothed from shoulder to toe. No shorts, short skirts, sandals or shoes that expose any part of the foot are permitted. It is recommended that you purchase and use a full-length lab coat. This will adequately cover the upper body, but your legs, ankles and feet must be covered by your 'street clothing'.

Laboratory reports: Post-laboratory assignments will be due at the start of the tutorial in the week shown on the course schedule. Post-lab assignments are to be handwritten in the laboratory notebook. Failure to submit a post-lab assignment for an experiment will result in zero credit for that experiment (no credit will be given for the pre-lab or data & observations parts in the absence of a full report.) There will be one formal (word-processed) laboratory report that will be worth 35 points. Details on this report are in the laboratory manual.

Adjustments to laboratory scores: The instructor will make every effort ensure that the grading of laboratory reports is consistent and fair. To this end, the instructor reserves the right to normalize the laboratory scores from the different laboratory instructors to the same average. Any such adjustment will be made at the end of the semester after all scores have been submitted. TA performance will be assessed throughout the semester with the goal of eliminating any necessity for these adjustments. Students are encouraged to bring any concerns about the equity of the grading process to the attention of the course instructor.

CLASS POLICY ON LATE (OR EARLY) ASSIGNMENTS:

Laboratory reports: Late laboratory reports will be penalized by the loss of 20% of the total points per day (or portion thereof) that they are late. *Reports submitted after the start of tutorial are a day late!* This penalty is applied after the normal grading of the report. Reports submitted more than one week late will receive zero points. No reports will be accepted after 5:00 pm on the last day of classes (Friday, April 29, 2016) even if they are not yet one week late.

Homework assignments, Reading assignments, & In Lecture assignments: No extensions to the due times of assignments will be granted for any reason. There will be no opportunity to make-up the points on any of these assignments once its due time has passed.

Early submission: If you know that you will not be present at the time a laboratory report is due, they may be submitted early without penalty. Homework assignments may be completed on the Mastering Chemistry as soon as the homework assignment is posted.

Method of submission: It is best to personally deliver late or early submissions to the instructor or TA. Note that, outside of class/laboratory times and posted office hours, we make no pledge to be present or available for this purpose. If you are submitting work at other than the specified time, it is your responsibility to find us. Material may be submitted to Fulmer 319A during normal office hours. Assignments delivered in any other way (slid under the instructor's office door, for example) will be considered to have been submitted at the time they are found, if they are found.

ACADEMIC INTEGRITY: Cheating or plagiarism in any form will not be tolerated. Cheating includes, but is not limited to: copying work or allowing your work to be copied; use of unauthorized material at quizzes and exams, any communication between students during a quiz or exam, and actively looking at another student's paper during a quiz or exam. Students repeating the course must rework and rewrite all assignments. Plagiarism includes resubmitting previously graded homework or lab reports from a previous semester, even if they were your own work. Plagiarism also includes using laboratory data from another person or a previous semester. Obtaining information about quizzes taken in other sections is considered cheating. Use of any electronic device other than an approved calculator during a quiz or examination is cheating. Students who violate WSU's Academic Integrity Policy (identified in Washington Administrative Code (WAC) 504-26-010(3) and -404) will receive a score of zero for that assignment, quiz or exam for the first incident. A second incident of cheating will result in an F for the course and possible dismissal from the University. All incidences are reported to the Office of Student Conduct and additional university sanctions may incur.

ACCOMODATIONS: Reasonable accommodations are available for students who have a documented disability. If you have a disability and may need accommodations to fully participate in this class, please visit or call the Access Center (Washington Bldg, Room 217, 509-335-3417) to schedule an appointment with an Access Advisor. All accommodations **MUST** be approved through the Access Center

Accommodations are available for students for whom examinations fall on days objectionable due to religious beliefs. Requests for such accommodation must be presented, in writing, to the course instructor at least one week prior to the examination.

EMERGENCY NOTIFICATION SYSTEM: "Washington State University is committed to enhancing the safety of the students, faculty, staff, and visitors. It is highly recommended that you review the Campus Safety Plan (<http://safetyplan.wsu.edu/>) and visit the Office of Emergency Management web site (<http://oem.wsu.edu/>) for a comprehensive listing of university policies, procedures, statistics, and information related to campus safety, emergency management, and the health and welfare of the campus community."

STUDENT LEARNING OUTCOMES: At the end of the course the student should

1. Have an understanding of the concepts, models, and theories that form the foundation of the field of chemistry
2. Apply the standard algorithmic calculation procedures, individually and in combination, associated with these concepts, models, and theories.
3. Be able to describe, explain, and predict the behavior and interactions of substances on the atomic, molecular and macroscopic levels.
4. Be able to communicate in the basic vocabulary of chemistry including the ability to transition between chemical names and chemical formula in a facile manner and the ability to describe reactions using balanced chemical equations.
5. Understand the relationship between molecular structure the physical and chemical properties of a substance.
6. Create procedures to solve problems by applying single and multiple concepts to new situations.
7. Apply chemical procedures and evaluate experimental results to develop an appreciation for the experimental basis of chemical knowledge and experimental methods through laboratory work, with special emphasis on qualitative analysis.
8. Write effectively about scientific experiments by describing laboratory procedures and results from both the student's laboratory experience and articles from the scientific literature. Be able to evaluate and present a discussion of these results in the manner of a scientific report.

ASSESSMENT: SLO #s 1-7 will be assessed using examinations, tutorial quizzes, laboratory experiments and reports, homework, in-lecture assignments, and a research paper. SLO #8 will be assessed using the quizzes, lab reports, and the research paper.

SCHEDULE

CHEMISTRY 101

SPRING 2017

	Date	Chapter	Topic	Lab Expt / Pre-lab due	Lab report due	Quiz/ Exam
1	Jan 9-13	1.1-1.6 2.1-2.3	Physical states, elements, compounds, chemical formulae, atoms, atomic nuclei.	Tutorial only.	None	None
2*	Jan 16 – 20	2.4-2.9 3.1-3.5	Electron configurations, Ions & ionic bonds, nomenclature.	Worksheet 1: <i>Calculations and units.</i>	Worksheet 1	Quiz 1
3	Jan 23 – 27	3.6-3.11, 4.11, 4.1-4.5	Nomenclature, acids & bases, covalent bonds, bond polarity, molecular compounds.	Worksheet 2: <i>Inorganic Nomenclature</i>	Worksheet 2	Quiz 2
4	Jan 30 – Feb 3	4.6-4.10	Lewis structures, VSEPR, shape, geometry, molecular polarity, the periodic table.	Experiment 1: <i>Molecular Models</i>	None	Quiz 3
5	Feb 6 – 10	1.7-1.12	Measurements, units, calculations, uncertainty, density energy, specific heat,	Tutorial plus Exam 1 practice	Experiment 1	Exam 1
6	Feb 13 – 17	5.1-5.7	Chemical equations, Reaction classes, oxidation numbers, net ionic equations	Experiment 4: <i>Laboratory Techniques</i>	None	Quiz 4
7*	Feb 20 – 24	6.1-6.4	The mole, stoichiometry.	Experiment 2: <i>Reactions & Equations</i>	Experiment 4	Quiz 5
8	Feb 27 – Mar 3	6.5, 7.1-7.3	Limiting reactant, heat of reaction, entropy, free energy.	Experiment 3: <i>Acid base titration</i>	Experiment 2	Quiz 6
9	Mar 6 - 10	7.4-7.6	Reaction rates.	Tutorial plus Exam II practice	Experiment 3	Exam 2
	Mar 13 - 17	SPRING BREAK				
10	Mar 20 - 24	7.7-7.9, 8.1-8.10	Chemical equilibrium, the gas laws, intermolecular forces.	Experiment 6: <i>Heat of Reaction</i>	None	Quiz 7
11	Mar 27 - 31	8.11-8.14, 9.1-9.8	Physical phases and phase changes, solutions, solubility, concentration units, electrolytes,	Experiment 7: <i>The molecular mass of a volatile liquid</i>	Experiment 6	Quiz 8
12	Apr 3 - 7	9.9-9.11, 10.1-10.6	Colligative properties, osmosis, acids and bases, conjugate acid/base pairs, acid/base equilibrium, the pH scale.	Experiment 8: <i>Colligative Properties</i>	Experiment 7	None
13	Apr 10 - 14	10.8-10.11	Acid/base reactions, buffers, titrations.	Tutorial plus Exam 3 practice	Experiment 8	Exam 3
14	Apr 17 - 21	11.1-11.7	Nuclear decay, nuclear reactions, half-life.	Experiment 5: <i>Spectrophotometry</i>	None	Quiz 9
15	Apr 24 - 28	11.8-11.9	Radiation, fission and fusion. Review.	Tutorial only	Experiment 5	Quiz 10
	May 1 (Monday)	<i>FINAL EXAM 7 P M - 1 0 P M</i>				

*Monday Holiday. No lecture on January 16, February 20.

Enter your Blackboard course

1. Sign in to Blackboard and enter your Blackboard course.
2. Do one of the following:
 - > Select any Pearson link in the Content area.
 - > Select **Tools** in the left navigation and **Pearson's MyLab & Mastering** on the Tools page. Next, select any course link in the top area of the Pearson's MyLab & Mastering Tools page.

Get access to your Pearson course content

1. Enter your Pearson account **username** and **password** to **Link Accounts**.
You have an account if you have ever used a Pearson MyLab & Mastering product, such as MyMathLab, MyITLab, MySpanishLab, MasteringBiology or MasteringPhysics.
 - > If you don't have a Pearson account, select **Create** and follow the instructions.
2. Select an access option:
 - > Enter the access code that came with your textbook or was purchased separately from the bookstore.
 - > Buy access using a credit card or PayPal account.
 - > If available, get temporary access by selecting the link near the bottom of the page.
3. From the You're Done page, select **Go to My Courses**.

Note: We recommend you always enter your MyLab & Modified Mastering course through Blackboard.

Get your computer ready

For the best experience, check the system requirements for your product at:
<http://www.pearsonmylabandmastering.com/system-requirements/>

Need help?

For help with MyLab & Modified Mastering with Blackboard, go to:
<http://help.pearsoncmg.com/mylabmastering/bbi/student/en/index.html>