

# SYLLABUS

# CHEMISTRY 105

# SPRING 2015

**LECTURES:** MWF 11:10AM and 2:10PM

Fulmer 226

**INSTRUCTOR:** Dr. Michael Finnegan Fulmer 30A 335-5692 *email: mgfinnegan@wsu.edu*

Office Hours: 10:00AM-11:00AM MWF, 2:00-3:00PM Tu/Th.

**GENERAL CHEMISTRY OFFICE:** Nikki Clark Fulmer 319A 335-1516 [nikki\\_clark@wsu.edu](mailto:nikki_clark@wsu.edu)

**LABORATORY SUPERVISOR:** Ryan Rice Fulmer 313 335-6358 [rwrice@wsu.edu](mailto:rwrice@wsu.edu)

<b>GRADING:</b>	3 "midterm" exams	225	<b>GRADE RANGES:</b> (minimum points to achieve)		
	11 tutorial quizzes (best 8)	120	900 points	A	740 points C+
	Homework	85	870 points	A-	700 points C
	Reading Assignments	80	840 points	B+	670 points C-
	In Lecture Assignments	80	800 points	B	640 points D+
	10 laboratory experiments/worksheets	200	770 points	B-	600 points D
	Final Exam	<u>210</u>	Less than 600 points: F		
	<b>TOTAL</b>	<b>1000</b>			

**MIDTERM EXAMS:** Thursday Feb 12 6:00– 7:00 pm (Chapters 1-4+ Lab WS #1 & Expts 1 & 3)

Thursday March 12 6:00– 7:00 pm (Chapters 5-8.5 + Experiments 6, 7, & 9)

Thursday April 16 6:00– 7:00 pm (Chapters 8.6-10.5 + Experiments 4, 5, & 10)

**FINAL EXAM** Monday May 4 7:00pm–10:00 pm (Chapters 1-11 + all Worksheets/Experiments)

**TEXT:** *Principles of Chemistry: a Molecular Approach, 3<sup>rd</sup> Ed.* by Nivaldo J. Tro, Pearson education (2014).

ISBN: 978-0-321-80924-7 or 978-1-269-93260-8 (WSU custom cover). The text and access to the Mastering Chemistry homework site are required. The bookstores have new texts bundled with a modified Mastering Chemistry access code. Modified Mastering Chemistry access codes can also be purchased separately via the MasteringChemistry link on the blackboard course site.

**LAB TEXT:** *Chemistry 105-106 General Chemistry Laboratory Manual* by WSU Chemistry Department, Star Publishing (2014) is required to complete the laboratory portion of this course.

**LABORATORY NOTEBOOK:** Duplicating with numbered pages. (Sold in Fulmer 318 the 1<sup>st</sup> and 2<sup>nd</sup> week of class.)

**GOGGLES:** Required by State Law. (Sold in Fulmer 318 the 1<sup>st</sup> and 2<sup>nd</sup> week of class.)

**LABORATORY COAT:** Optional but recommended. A strict dress code is enforced in the laboratories. NO SHORTS, NO SHORT SKIRTS, NO SANDALS, NO BARE MIDRIFTS. (See laboratory dress code.)

**CALCULATORS:** You are expected to have and to be able to use a scientific calculator. Graphing calculators are allowed but not required. The use of any stored information/programs in a programmable calculator will be considered cheating. Calculators with a full QWERTY keyboard (such as the TI-92, TI-nspire or Voyage 200); PDAs; palmtop, laptop and handheld computers; and cell phone/calculator combinations may not be used during quizzes and examinations. You are responsible for bringing your calculator to all tutorials, lectures, labs and exams.

**ONLINE COMPONENTS:** We will be using Blackboard Learn for course management and online information. This can be accessed via <https://learn.wsu.edu>. Use your WSU network ID and password to log in. All online aspects of the course including homework sets (MasteringChemistry), reading assignments (MasteringChemistry), and In Lecture Assignments (Learning Catalytics) should be accessed via the links on the Blackboard site. All e-mail communications to the course instructor and TAs should be via the Blackboard mail tool. Confidential information such as scores and grades may not be transmitted via unsecured email.

**LECTURES:** Lectures must be attended on a regular basis. You will be expected to read the textbook AHEAD of coming to class. Lectures will supplement and clarify the information from your text rather than reiterate it. Lectures will focus on problem solving and demonstrations of chemical reactions. There will be in-lecture problem assignments via the LearningCatalytics website (accessible via the course Blackboard site). These will be graded and counted towards your total point score. These in-lecture assignments will be unannounced, cannot be made up. *Bring a calculator and web-enabled device to all lectures.* You are encouraged to form collaborative study groups and to sit with your group members during lecture.

**READING ASSIGNMENTS:** There will be reading assignments due before each lecture starting with the Wednesday, January 14<sup>th</sup> lecture (lecture #1). These consist of assigned sections of the text and questions intended to evaluate your understanding of this material. The reading assignments are available on the MasteringChemistry site which should be accessed via the Blackboard course site. Each assignment must be completed by 2:00AM on the day of the associated lecture. Each reading assignment will be available a week before they are due (except reading assignments 1 and 2, which will be available on Monday, January 12<sup>th</sup> and are due at 2:00AM on 1/14/15 and 1/16/15 respectively). Each reading assignment is worth 3 point. Your score will be determined by multiplying your percent correct by 3 and rounding to the nearest ½ point. 123 points will be available from the reading assignments this semester (41 reading assignments × 3 points). A maximum of 80 points from the reading assignments will count toward the course grade. A portion of any reading assignment points obtained in excess of 80 points might be counted as extra credit at the end of the semester. It is important to note that the completion of these assignments is independent of lecture attendance. If you are sick or out of town, it is still possible to complete the assignments.

**IN-LECTURE ASSIGNMENTS:** Most lectures will include problem assignments via the LearningCatalytics website (accessible via the course Blackboard site). These assignments will be identified as “In-Lecture Assignments” in the grade book. These sessions are interactive and will require a Wi-Fi enabled device such as a smart-phone, laptop, or tablet. Each in-lecture assignment will be worth 4 course points and will be graded on both participation (25%) and correctness of answers (75%). There will be in excess of 150 points available from in-lecture assignments. A maximum of 80 points from the in-lecture assignments will count toward the course grade. A portion of any in-lecture assignment points obtained in excess of 80 points might be counted as extra credit at the end of the semester. Each in-lecture assignment will be available only during a portion of the associated lecture. There will be no make-up opportunities for in-lecture assignments. If you are not present for lecture for any reason, you will not have the opportunity to get that 4 points. If your Wi-Fi device fails during the assignment, you will not have an opportunity to get the points for the portion of the assignment you are not able to complete. Make certain that the batteries are fully charged before arriving for lecture.

**HOMEWORK:** There will be weekly homework assignments consisting of a minimum of ten questions each. These assignments are administered by Mastering Chemistry and accessed through the course Blackboard site. You should have an access code for Mastering Chemistry bundled with your text. Please follow the attached instructions and use the access code to register for the system. If you purchased a text that did not include an access code, follow the attached instructions to purchase a code and register for the system.

A new homework assignment will be made available each week (no later than 5:00PM each Monday). Each assignment must be completed by 8:00PM the following Monday in order to obtain full credit. Partial credit may be obtained up to 5 days after the due time (available points decrease by 20% each day). The due date/time for each assignment will be listed with the assignment on the homework site. Each homework set will be pro-rated to have a value of 8 course points. Thus there will be 120 homework points available this semester. A maximum of 85 points from homework will count toward the final grade. A portion of any homework points obtained in excess of 85 points might be counted as extra credit at the end of the semester.

Each homework assignment has an **Adaptive Follow-Up** assignment. These assignments are customized for you and provide additional practice on homework problems with which you had difficulty. If you score 90% or higher on the homework assignment, you are automatically assigned full credit for the Adaptive Follow-up, once you open it. If you score below 90%, you must complete the Follow-up assignment to be awarded these points.

**TUTORIALS:** These are small classroom meetings associated with your laboratory section and led by your TA. Students who miss tutorial will **not** be allowed into the lab. Quizzes are given in tutorial most weeks (see the course schedule). Tutorials are interactive problem solving sessions driven by your questions. Bring your text, lab manual and calculator to tutorial. Pre-labs and lab reports are due at the start of tutorial. Help with pre-labs and post-labs will not be available in tutorial as they must be completed before attending tutorial. **Tutorial sessions are never canceled!** If your TA fails to arrive for a tutorial section, send one person to contact Dr. Finnegan or Ryan Rice immediately. All others must remain in the tutorial room until the TA or a substitute arrives. Students who leave tutorial under these circumstances will forfeit all points associated with that tutorial/laboratory session (lab report, lab, and quiz).

**QUIZZES:** There will be eleven 15-point quizzes of which the best eight will count. Quizzes are given in tutorial. Quizzes will cover lecture, homework and laboratory material. You will be allowed to prepare a single 3" × 5" card containing your **HAND-WRITTEN** notes for use during each of the quizzes. No other handwritten material and no printed or photocopied material may be used during the quiz, except for an approved periodic table (the table that appears on the back of your laboratory manual).

**LABORATORIES:** Your course section includes a lecture time and a laboratory time. This is a laboratory GER course, thus the laboratory must be completed in order to pass the course. Thus, obtaining a score of zero for 3 or more experiments will result in an F for the course.

**Make-up labs:** Labs missed for reasons beyond your control, may be made up, on a space available basis, in the same week that the lab is missed. You will be allowed to make up a maximum of two labs per semester in this manner. Permission for a make-up lab must be obtained, in writing, from the Chemistry Office, Fulmer 319A. The permission slip will be collected and signed by the make-up TA. **We cannot guarantee that make-up space will be available.** If you know in advance that you will miss a lab, visit Fulmer 319A as soon as possible in order to maximize the chance that make-up space will be available. **If make-up space is not available:** Bring your completed pre-laboratory assignment to Dr. Finnegan's office (Fulmer 30A) to be supplied with make-up data for the scheduled experiment. Do this as soon as you can! Reports based on make-up data are due at the normal time (in tutorial one week after you should have attended lab) and will be worth no more than ½ credit.

**Pre-laboratory assignments:** Pre-laboratory assignments are due at the start of the tutorial. Students who fail to submit a complete pre-lab assignment at this time will be assessed a late penalty on the full report and be required to complete the pre-lab assignment before they are admitted to lab. The student will not be given extra time in the laboratory to make up for laboratory time spent completing the pre-lab.

**Laboratory procedure:** Students are to perform the experiments individually unless the laboratory manual specifically requires partners for the experiment being performed. Each student is expected to record all data and observations for each experiment directly into their own laboratory notebook. Data may not be recorded on loose, 'scratch' paper then transferred to the notebook. Submission of identical data by two or more students who are not assigned to be laboratory partners will be considered cheating. Appropriate penalties will be applied to all parties. Some experiments will require you to transfer data from your notebook into a laboratory computer before you leave lab. This data must remain in the notebook! You are required to get your TA's signature on your data/observations at the end of the experiment. You will then submit the original copy of the data to your TA before you leave lab.

**Laboratory dress code:** For your safety, a strict dress code will be enforced in the laboratory. Failure to comply with the dress code will result in expulsion from the laboratory and a consequent score of zero for that experiment. The dress code requires that you be fully clothed from shoulder to toe. No shorts, short skirts, or shoes that do not cover the entire foot are permitted. It is recommended that you purchase and use a full-length lab coat. This will adequately cover the upper body, but your legs, ankles and feet must be covered by your 'street clothing'.

**Post-laboratory reports:** Laboratory reports (post-labs) will be due at the start of the tutorial in the week shown on the course schedule. Failure to submit a laboratory report for an experiment will result in zero credit for that experiment (no credit will be given for the pre-lab or data & observations sections in the absence of a full report.)

**Late submission penalties:** Late laboratory reports will be penalized by the loss of 20% of the total points per day (or portion thereof) that they are late. *Reports submitted after the start of tutorial are a day late!* This penalty is applied after the normal grading of the report. Late penalties are applied to the entire experiment, not just the portion of the report that is late. Late penalties assessed for different parts of the report are cumulative. Reports submitted more than one week late will receive zero points. No reports will be accepted after 5:00 pm on the last day of classes (Friday, May 1<sup>st</sup>, 2015) even if they are not yet one week late.

**Early submission:** If you know that you will not be present at the time a laboratory report is due, they may be submitted early without penalty.

**Method of submission:** It is best to personally deliver late or early submissions to the instructor or TA. Note that, outside of class/laboratory times and posted office hours, we make no pledge to be present or available for this purpose. If you are submitting work at other than the specified time, it is your responsibility to find us. Material may be submitted to Fulmer 319A or Fulmer 305 during normal business hours (8:00AM-5:00PM M-F). Assignments delivered in any other way (slid under the instructor's office door, for example) will be considered to have been submitted at the time they are found, if they are found.

**Adjustments to laboratory scores:** The instructor will make every effort ensure that the grading of laboratory reports is consistent and fair. To this end, the instructor reserves the right to normalize the laboratory scores from the different laboratory instructors to the same average. Any such adjustment will be made at the end of the semester after all scores have been submitted. TA performance will be assessed throughout the semester with the goal of eliminating any necessity for these adjustments. Students are encouraged to bring any concerns about the equity of the grading process to the attention of the course instructor.

**EXAMS:** There will be three midterm exams and a comprehensive final. All exams will be multiple-choice. You will be responsible for bringing a calculator and a pencil to all exams. A bubble-in answer sheet will be provided. No notes or books are allowed. Exams may be given in rooms other than the regular classroom. These rooms will be announced. No make-up exams will be given. If you are unable to take a scheduled midterm exam for academic reasons beyond your control, you will be allowed to schedule the exam at an earlier time. Midterm exams missed due to illness will be excused, with the other exams plus the final pro-rated to count for more. **Evening exams take precedence over all other university activities.**

**ACADEMIC INTEGRITY:** Cheating or plagiarism in any form will not be tolerated. Cheating includes, but is not limited to: copying work or allowing your work to be copied; use of unauthorized material at quizzes and exams, any communication between students during a quiz or exam, and actively looking at another student's paper during a quiz or exam. Students repeating the course must rework and rewrite all assignments. Plagiarism includes resubmitting previously graded homework or lab reports from a previous semester, even if they were your own work. Plagiarism also includes using laboratory data from another person or a previous semester. Obtaining information about quizzes taken in other sections is considered cheating. Use of any electronic device other than an approved calculator during a quiz or examination is cheating. All incidences of cheating will be reported to the Office of Student Affairs. The first incidence of cheating will result in a score of zero for that assignment, quiz or exam. A second incident of cheating will result in an F for the course and possible dismissal from the University.

**ACCOMODATIONS:** Reasonable accommodations are available for students who have a documented disability. If you need accommodations to fully participate in this class, please visit the Access Center. All accommodations **MUST** be approved through the Access Center (Washington Bldg, Room 217). Please stop by or call 509-335-3417 to make an appointment with an Access Advisor. Further information is available at <http://accesscenter.wsu.edu>

Accommodations are available for students for whom examinations fall on days objectionable due to religious beliefs. Requests for such accommodation must be presented, in writing, to the course instructor at least one week prior to the examination.

**EMERGENCY NOTIFICATION SYSTEM:** Washington State University is committed to enhancing the safety of the students, faculty, staff, and visitors. It is highly recommended that you review the Campus Safety Plan (<http://safetyplan.wsu.edu/>) and visit the Office of Emergency Management web site (<http://oem.wsu.edu/>) for a comprehensive listing of university policies, procedures, statistics, and information related to campus safety, emergency management, and the health and welfare of the campus community.

**STUDENT LEARNING OUTCOMES:** Chemistry 105 fulfills three credits of Inquiry in the Physical Sciences [PSCI] and one credit of laboratory as part of the WSU Common Course Requirements (UCORE). As with all UCORE courses, Chemistry 105 is designed to advance students toward the WSU Learning Goals, especially Scientific Literacy, Critical and Creative Thinking, Quantitative Reasoning, and Information Literacy. In particular, students who successfully complete Chemistry 105 will be able to:

1. Develop an understanding of the concepts, models, and theories that form a foundation for the field of chemistry (the understanding of how the behavior of matter is determined by the properties of atoms and molecules).
2. Remember the basic vocabulary of chemistry, the metric prefixes and the names of the most common elements.
3. Apply standard algorithmic calculation procedures, individually and in combination, that relate macroscopic properties, including mass, volume, pressure, and temperature of substances. Be able to balance chemical reactions and relate amounts of reactants and products as well as associated energy changes. In addition, be able to relate macroscopic and atomic level properties of numbers of atoms and molecules, chemical formulas, and molecular structures and properties.
4. Apply models of bonding to predict and describe the structure of molecules including their physical properties.
5. Create procedures to solve problems by applying single and multiple concepts to new situations.
6. Apply chemical procedures and evaluate experimental results to develop an appreciation for the experimental basis of chemical knowledge and experimental methods through laboratory work. (Lab reports)
7. Write effectively about scientific experiments by describing laboratory procedures and results, and then evaluating and presenting a discussion of these results in the manner of a scientific report.

**ASSESSMENT:** The SLOs 1 through 5 will be assessed with reading, in-class, and homework assignments done through an online system. In addition they will be assessed with tutorial quizzes and the evening exams. SLOs 6 and 7 will be assessed through the written lab reports.

**SCHEDULE****CHEMISTRY 105****SPRING 2015**

	Date	Chapter	Topic	Lab Expt / Pre-lab due	Post-lab due	Quiz/ Exam	
1	Jan 12–16	1, 2, 3.1-3.6	Atoms, ions, atomic mass, the mole, bonds, chemical formulae & nomenclature, acids.	Tutorial only.			
2*	Jan 19 – 23	3.7-3.11 4.1-4.2	Molar mass, empirical formula, chemical equations, stoichiometry	Worksheet 1: <i>Inorganic Nomenclature</i> .	Worksheet 1	Quiz 1	
3	Jan 26–30	4.3-4.7	Limiting reactants, solution stoichiometry, precipitation reactions.	Experiment 1: <i>Laboratory Techniques</i>		Quiz 2	
4	Feb 2–6	4.8-4.9	Acid-base reactions, net ionic equations, redox reactions, oxidation numbers.	Experiment 3: <i>Acids and Bases</i>	Experiment 1	Quiz 3	
5	Feb 9–13	5.1-5.9	Gasses: properties, stoichiometry, kinetic molecular theory, partial pressure.	Tutorial plus Exam 1 practice	Experiment 3	Exam 1	
6*	Feb 16–20	6.1-6.6	Thermodynamics, specific heat, enthalpy, thermochemistry, calorimetry.	Experiment 9: <i>Preparation of an Iron Oxalate Complex</i>		Quiz 4	
7	Feb 23–27	6.7-6.9 7.1-7.3	$\Delta H$ , Hess's Law. Electromagnetic radiation, wave-particle duality, atomic spectra.	Experiment 7: <i>The Density of Air</i>	Experiment 9	Quiz 5	
8	Mar 2–6	7.4-7.6 8.1-8.5	The quantum mechanical atom, orbitals, electron configurations, valence electrons.	Experiment 6: <i>Heat of formation of ammonium chloride</i>	Experiment 7	Quiz 6	
9	Mar 9–13	8.6-8.9	Periodic trends, magnetic properties.	Tutorial plus Exam 2 practice	Experiment 6	Exam 2	
	Mar 16– 20	<b>SPRING BREAK</b>					
10	Mar 23–27	9.1-9.4, 9.10-9.11	Ionic bonds, lattice energies, covalent bonds, bond energy, bond length, metallic bonds.	Experiment 4: <i>Molar mass of a known acid</i>		Quiz 7	
11	Mar 30– Apr 3	9.5-9.9	Lewis structures, resonance, formal charge.	Experiment 5: <i>Molar mass of an unknown acid</i>	Experiment 4	Quiz 8	
12	Apr 6–10	10.1-10.5	VSEPR theory, molecular shape and polarity	Experiment 10: <i>Analysis of bleach by Iodometry</i>	Experiment 5	Quiz 9	
13	Apr 13–17	10.6-10.8	Valence bond theory, molecular orbital theory	Tutorial plus Exam 3 practice	Experiment 10	Exam 3	
14	Apr 20–24	11.1-11.7	Intermolecular forces, physical properties, phase changes	Experiment 8: <i>The shapes of molecules and Ions</i>		Quiz 10	
15	Apr 27–May 1	11.8-11.13	Phase diagrams, Types of solids.	Tutorial only	Experiment 8	Quiz 11	
	May 4 (Monday)	<b>FINAL EXAM 7:00 PM - 10:00 PM</b>					

\*Monday Holiday: No lecture on January 19<sup>th</sup> or February 16<sup>th</sup>

## Get started with Modified MasteringChemistry and Blackboard Learn

### First, make certain that you have these 2 things...

**Email address:** You'll get important emails from your instructor at this address.

**Access code or credit card:** The required access code is part of the textbook bundle purchased at the bookstores. If you did not purchase the bundle, you can buy instant access with a credit card or a PayPal account during registration.

### Next, register!

1. Log in to Blackboard Learn, <https://learn.wsu.edu>, using your WSU network ID and password. Click on the link to our course: "Chemistry 105: Principles of Chemistry I (2015-SPRI-PULLM-CHEM-105-1260-LEC)". Click on the link to **Mastering Chemistry Course Home** to begin the registration process.
2. The registration page will appear in a new window or tab. It will start with the Mastering user agreement. Click the "I accept" button.
3. Sign into or create your Pearson account:  
You will be prompted to login with your Pearson account. If you have a Pearson account, enter the username and password. If you don't have a Pearson account, select the option to Create a new Pearson account. If you have ever used a Pearson Publishing on-line product before, you have a Pearson account. If you are not certain, select "Create" and enter the email address and username you want to use. The system will tell you if there is already an account associated with that email/user name.
4. The next page will offer you four options:
  - If you purchased the text bundle from the bookstore, click the button labeled "Access Code". Your access code is in a cardboard insert wrapped with the text book. Enter your **Access Code** in the spaces provided on the page that appears.  
-OR-
  - Purchase Mastering Chemistry access with the e-book text by clicking on the upper button in the **Use a Credit Card or PayPal** section (the right column). The cost is listed on the button. Follow the instructions to complete your purchase.  
-OR-
  - Purchase Mastering Chemistry access without the e-book by clicking on the lower button in the **Use a Credit Card or PayPal** section (the right column). The cost is less than for the homework/e-book combination. But, if you choose this option, you will be required to pay separately for access to LearningCatalytics (\$12 at the LearningCatalytics site). Follow the instructions to complete your purchase.  
-OR-
  - If you are not yet ready to purchase homework access, you may select the "Get temporary access without payment for 14 days". If you choose to do this, keep in mind that you will lose access to and credit for the homework if you do not actually purchase a code before the fourteen days are up.
5. You are now registered! Click on the "Go to your course" button to access your Modified Mastering product.