

Hydrogen and Power to Liquid (PtL) Concepts for SAF Production

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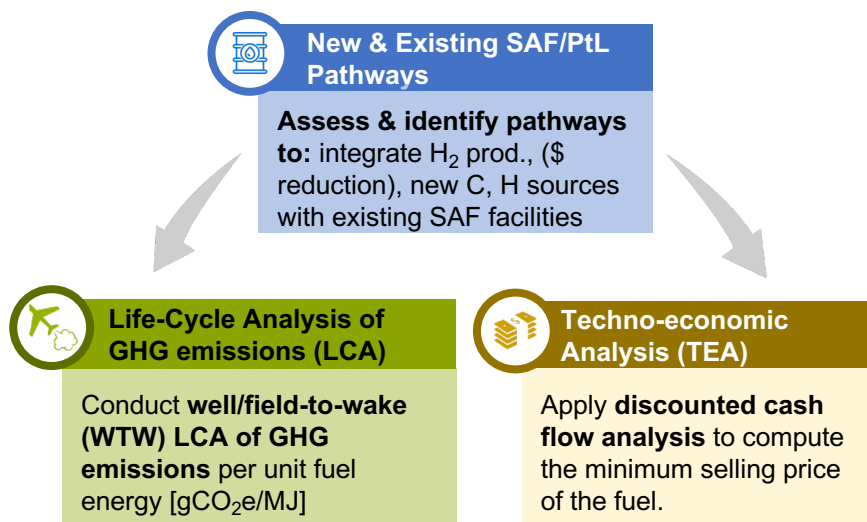
Objective:

Evaluate the opportunities associated with novel approaches for integrating hydrogen production and Power-to-Liquid (PtL) concepts into SAF production systems, including specifically lifecycle GHG emissions benefits and cost impacts.

Project Benefits:

1. Analysis of current and future SAF and PtL pathways: pros & cons, co-location potential
2. Provide a harmonized model to compare GHGs and costs of novel & existing pathways while capturing uncertainty and variability
3. Derive recommendations for future SAF/PtL R&D: how to best combine C, H, and conversion tech for lowest cost and GHG emissions

Research Approach:



Major Accomplishments (to date):

1. Defined electricity-based SAF pathways
2. Identified novel conversion pathways with renewable hydrogen use
3. Set up stochastic LCA and TEA framework for pathway analysis

Future Work / Schedule:

- Assessment of H₂ production, new C & energy sources with SAF
- LCA (gCO₂e/MJ) and TEA (\$/L) model
- First-principles-based analyses of DAC

Energy carrier vectors for aviation – a typology

● — Drop-in or near-drop-in SAF — ● ● — Non-drop-in energy carriers — ●

H
source

**Fossil
Jet-A**

Fossil H₂

**Waste- and
biomass-based
SAF**

Biomass and waste
streams; currently
supplemented by
H₂ produced from
SMR

**Power-and-
biomass-to-
Liquid (PBtL)**

Biomass and H₂
produced from
renewable
electricity

**Power-to-
Liquid**

H₂ produced from
low-carbon
electricity

Hydrogen

H₂ produced from
low-carbon
electricity or SMR
with CCS

**Battery
?**

Fossil carbon

Biomass and waste
streams*

Biomass

Atmospheric CO₂
or waste CO₂
streams

--

Atmosphere

Fossil fuel
combustion



Geosphere

Atmosphere

Photo-
synthesis

Biofuel
combustion

Biosphere

Atmosphere

Direct Air
Capture

Synfuel
combustion

Synthetic fuel

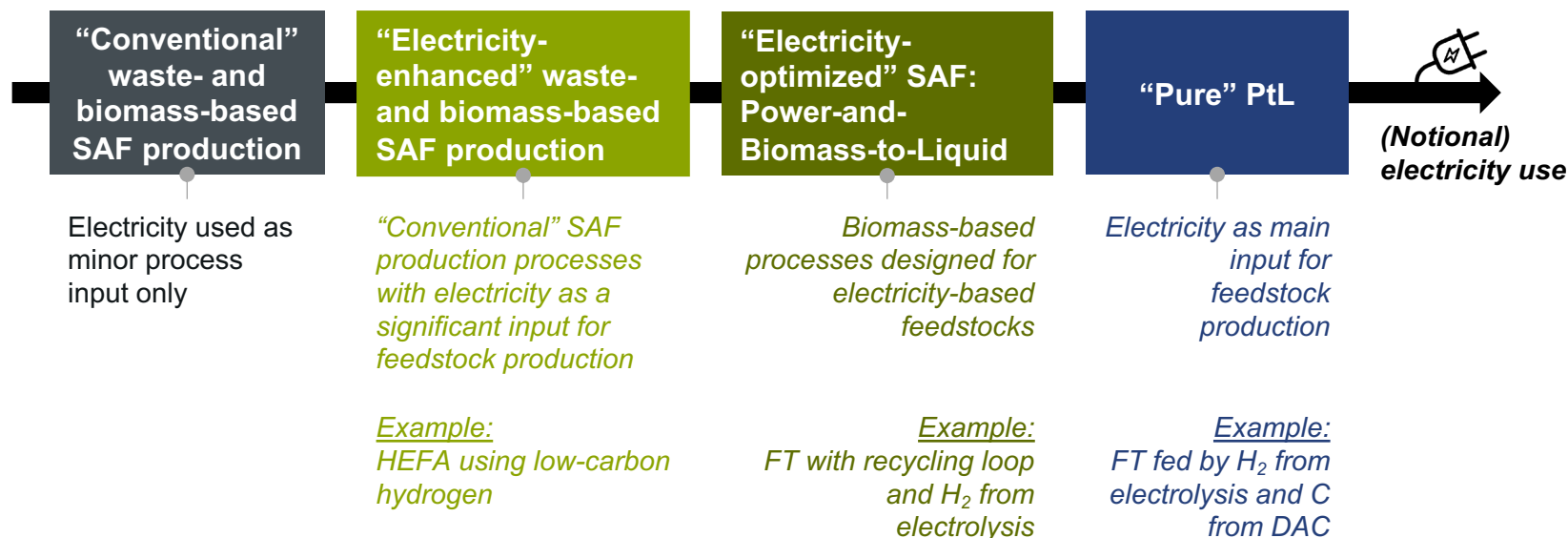
* Using carbon from waste streams provides a carbon benefit if the carbon content of the waste stream would have been released to the atmosphere anyways.

C source &
C balance

Mapping electricity use in SAF production – a significant opportunity space.

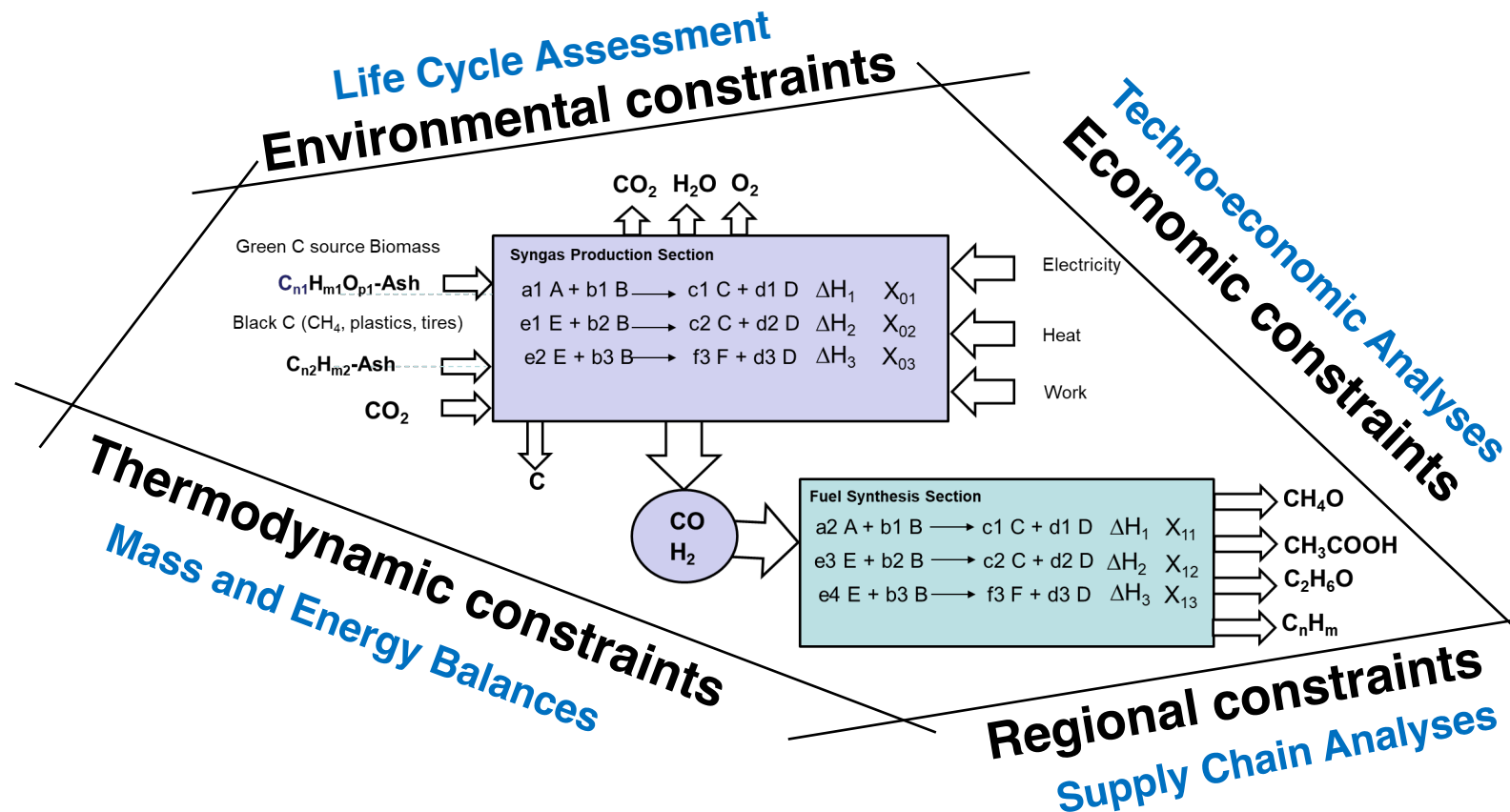
PtL fuels are a *class of synthetic drop-in hydrocarbon fuels*, which use electricity as a major input “feedstock”, especially for H₂ production, CO₂ extraction, and/or conversion into fuels.

BUT: A large range of pathway designs exists, including “hybrid” designs



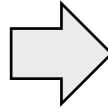
Objectives

Identify Carbon, Hydrogen, and Energy sources (New Potential Feedstocks) and fuel conversion processes to enhance the economics and carbon intensity of the SAF. Compare the environmental footprint and economic sustainability indicators of the most promising approaches with current SAF and conventional Power-to-Liquid pathways.

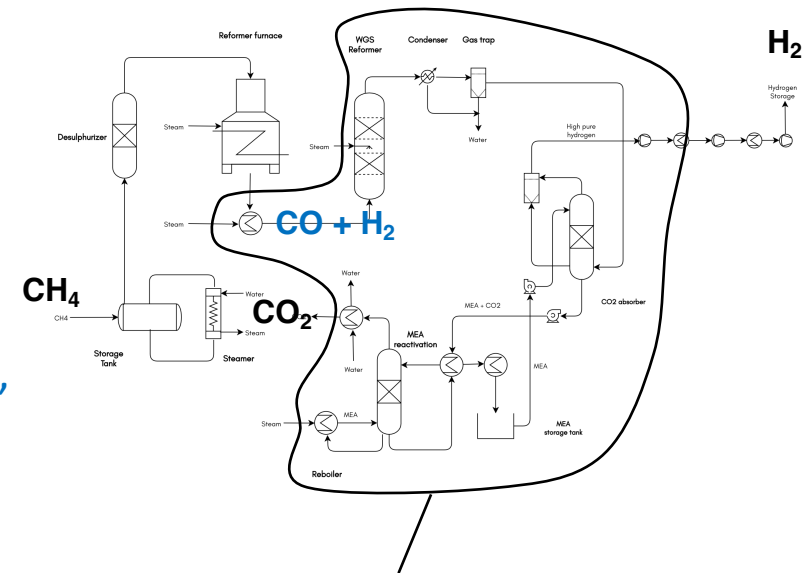


Review of new pathways to produce H₂

1. Steam Reforming (Low pressure, high pressure)
2. Dry Reforming
3. Auto-thermal Reforming
4. Methane Partial Oxidation
5. Thermal decomposition of hydrocarbons
6. Gasification of carbonaceous materials (biomass, coal, bitumen, MSW) with steam, CO₂ and O₂
7. Water Electrolysis (low and high temperature)
8. CO₂ Electrolysis



Steam Reforming



Section of an H₂ plant that may not be relevant when producing SAF

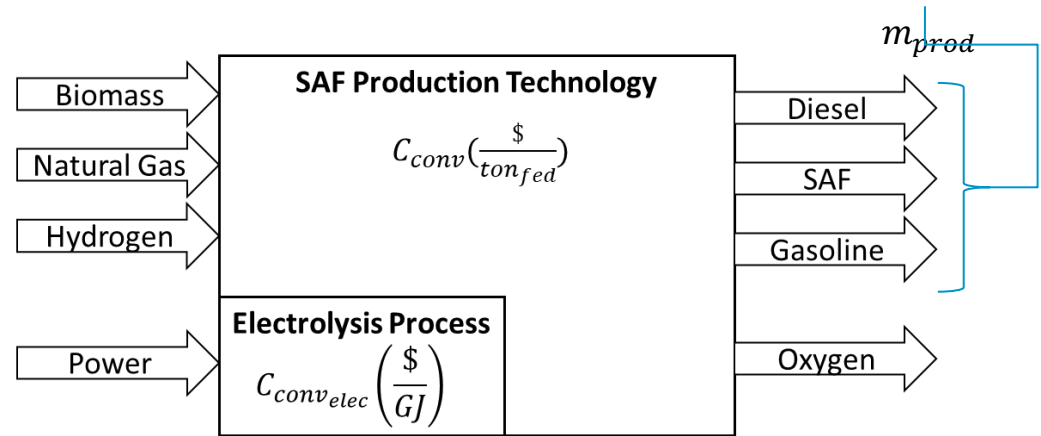
Goal: Write a Literature review on emerging and commercial hydrogen production technologies and techno-economic analysis of each of them.
The first draft is under review.

Technology-neutral framework

Lange has proposed simple models:

$$C_{prod} (\$/ton) \approx \frac{C_{feed} + C_{conv}}{yield}$$

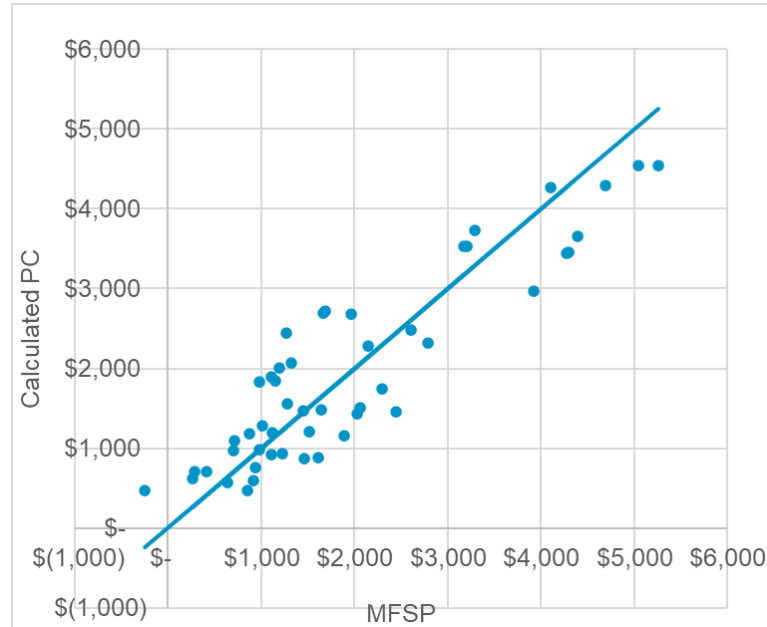
Resource Cost	Flowrate (tons)
$C_b = 70 \text{ \$}/\text{ton}$	m_b
$C_{NG} = 193 \text{ \$}/\text{ton}$	m_{NG}
$C_{H_2} = 4000 \text{ \$}/\text{ton}$	m_{H_2}
$C_{power} = 21.1 \text{ \$}/\text{GJ}$	$W_{elec}/m_{prod} \left(\frac{\text{GJ}}{\text{ton}} \right)$
$C_{O_2} = 40 \text{ \$}/\text{ton}$	m_{O_2prod}
$C_{prod} (\text{\$/ton})$	m_{prod}



$$C_{prod} = \frac{(C_b + C_{conv}) \cdot m_b + C_{NG} \cdot m_{NG} + C_{H_2} \cdot m_{H_2} - C_{O_2} \cdot m_{O_2prod}}{m_{prod}} + (C_{power} + C_{conv_{elec}}) \cdot \frac{W_{elec}}{m_{prod}}$$

Data selection and correlation

- 50 Datapoints from 8 studies
 - 28 used to fit base PC calculation
- Uncertainty of \$615/ton
 - Reasonable for economic analysis – 30% error the norm



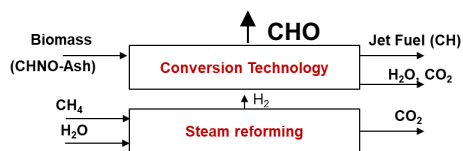
$$C_{prod} = \frac{(C_b + \$317) \cdot m_b + C_{NG} \cdot m_{NG} + C_{H_2} \cdot m_{H_2} - C_{O_2} \cdot m_{O_2prod}}{m_{prod}} + (C_{power} + 39) \cdot \frac{W_{elec}}{m_{prod}}$$

- \$317/ton consistent with values in chemical industry (100-300 \$/ton) (Lange, 2019)
- \$39/GJ approximates the levelized cost of electrolysis
- All other variables obtained from cost data

**The technology you use does not seem to be that important
What is vital, is the outcome of your technology**

Stoichiometric models

Current Commercial Technologies



Yield = **0.09 – 0.23**
 MFSP = **\$2050-5190/ton**

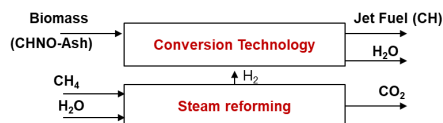
These very simple stoichiometric models consider that biomass O is removed preferentially from three molecules (CO_2 , H_2O , and O_2). The C left is converted into a hydrocarbon

All O removed as CO_2



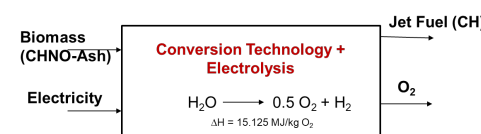
Yield = **0.34**
 PC = **\$1,195/ton**
 CI = **3.2 gCO2/MJ**
 RIN Effect = **\$605/ton**
 SPC = **\$590/ton**

All O removed as H_2O with external H_2



Yield = **0.55**
 PC = **\$762/ton**
 CI = **25 gCO2/MJ**
 RIN Effect = **\$393/ton**
 SPC = **\$369/ton**

All O removed as O_2 via electrolysis



Yield = **0.53**
 PC = **\$1,523/ton**
 CI = **11 gCO2/MJ**
 RIN Effect = **\$605/ton**
 SPC = **\$918/ton**


¹Yields Defined on biomass basis (tons distillate/tons biomass)

²CI calculated using WA average grid footprint (27 g/MJ)

PC: Production Cost

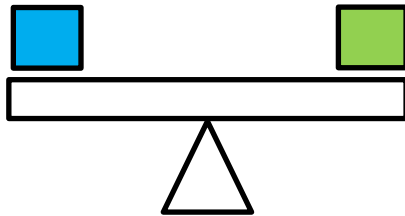
SPC: Subsidized production cost

Economic versus environmental trade-off

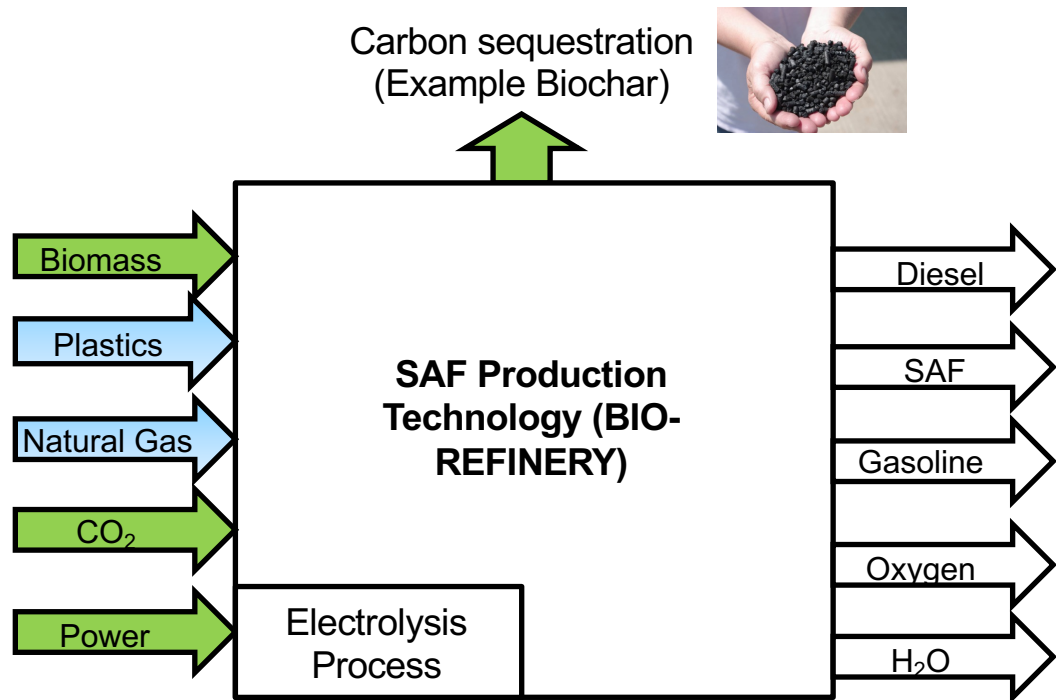
 Economic advantages

 Environmental advantages

Right Balance between
Economic and Environmental
competitive advantages?

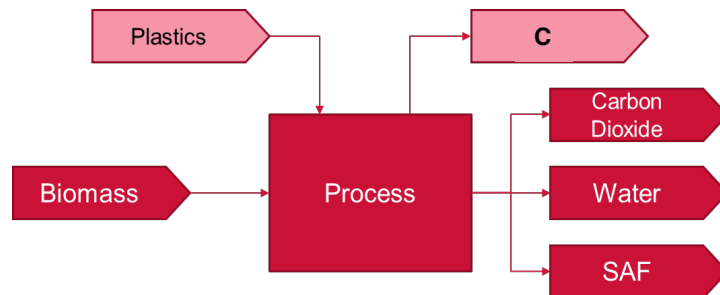


Balance will depend on **regulatory
frame + incentive levels**

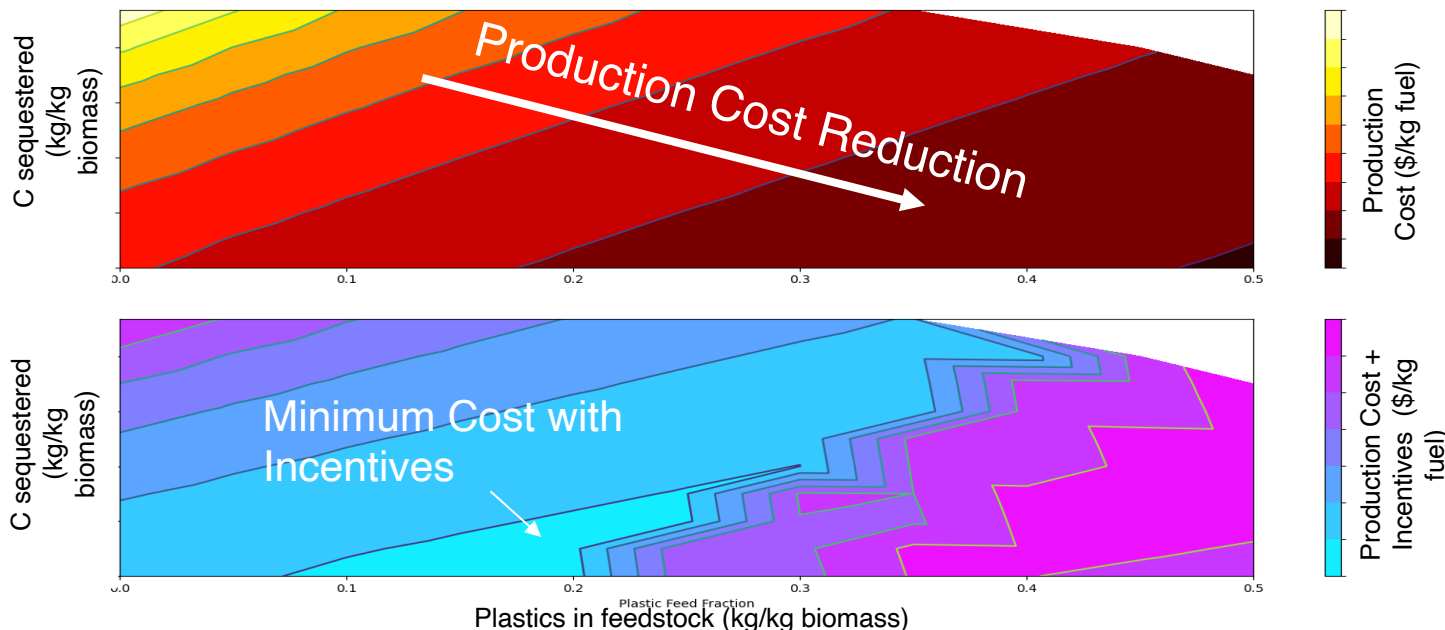


We will use the simplified models to study the trade-off between several technological alternatives

Removal of O in the form of CO₂



Plastics addition versus carbon sequestration

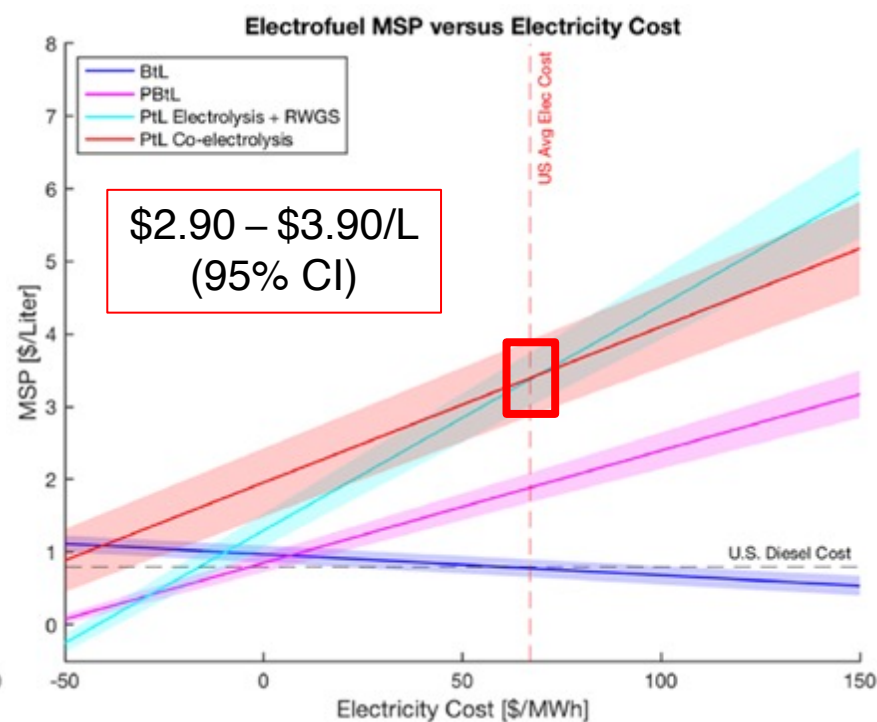
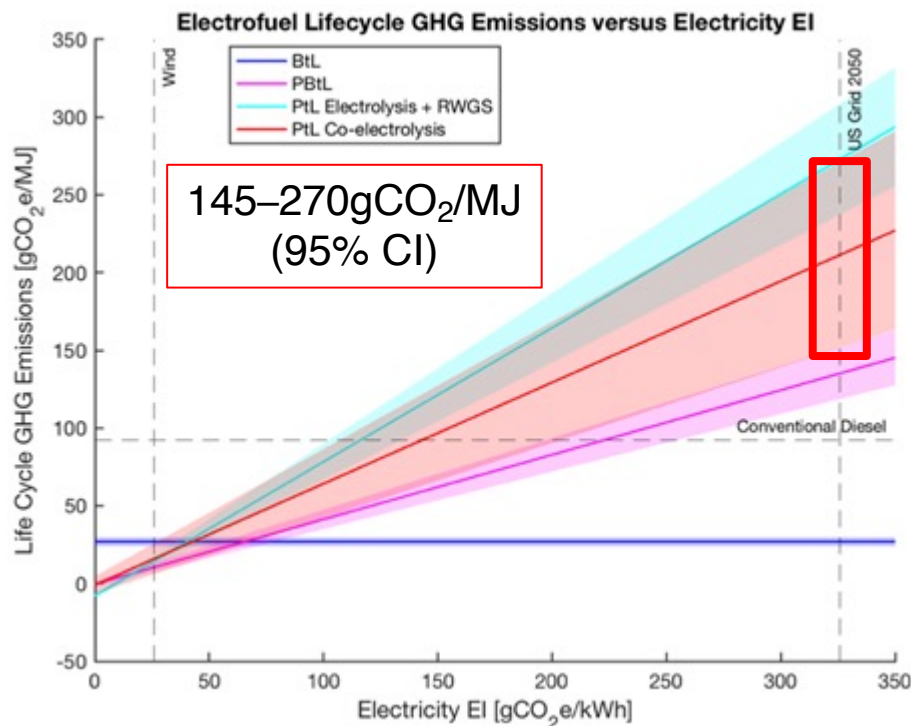


Carbon storage helps reduce the “carbon Intensity” of the fuel but reduces yield.

Plastics addition helps to increase yield but increases “carbon Intensity”

Under the incentives considered in our analysis (**RINs**, **LCFS**) the carbon sequestration benefits are insufficient to justify yield loss which directly impacts production cost. For the ideal model used there is run to add a very small quantity of plastics. This will increase the yield. Adding larger quantities will be penalized with the loss of incentives associated with environmental performance.

- There may be uncertainties in lifecycle GHG emissions and minimum selling prices of jet fuel for existing, new, and combinations of SAF/PtL pathways, especially since they are not yet used at scale.
- Stochastic assessment will allow us to understand the range of possible GHG emissions and costs of novel pathways, given prevailing uncertainties, thereby helping to identify opportunities and risks as well as guiding further R&D for pathway characterization.



Uncertainty

Refers to lack of data or an incomplete understanding of the factors that affect an outcome. More/better data could reduce uncertainty. (U.S. EPA, 2021). **Biofuel facility cannot intentionally control variable.**

Example

Feedstock transport mode share split (truck/rail/barge)

Primary focus in the context of analyzing novel unexplored technologies

Variability

Refers to inherent heterogeneity of diversity of data in an assessment. “A quantitative description of the range or spread of a set of values.” More/better data will not reduce variability (U.S. EPA, 2021). **Variable can be intentionally controlled or differs regionally.**

Example

Jet volume fraction in product slate

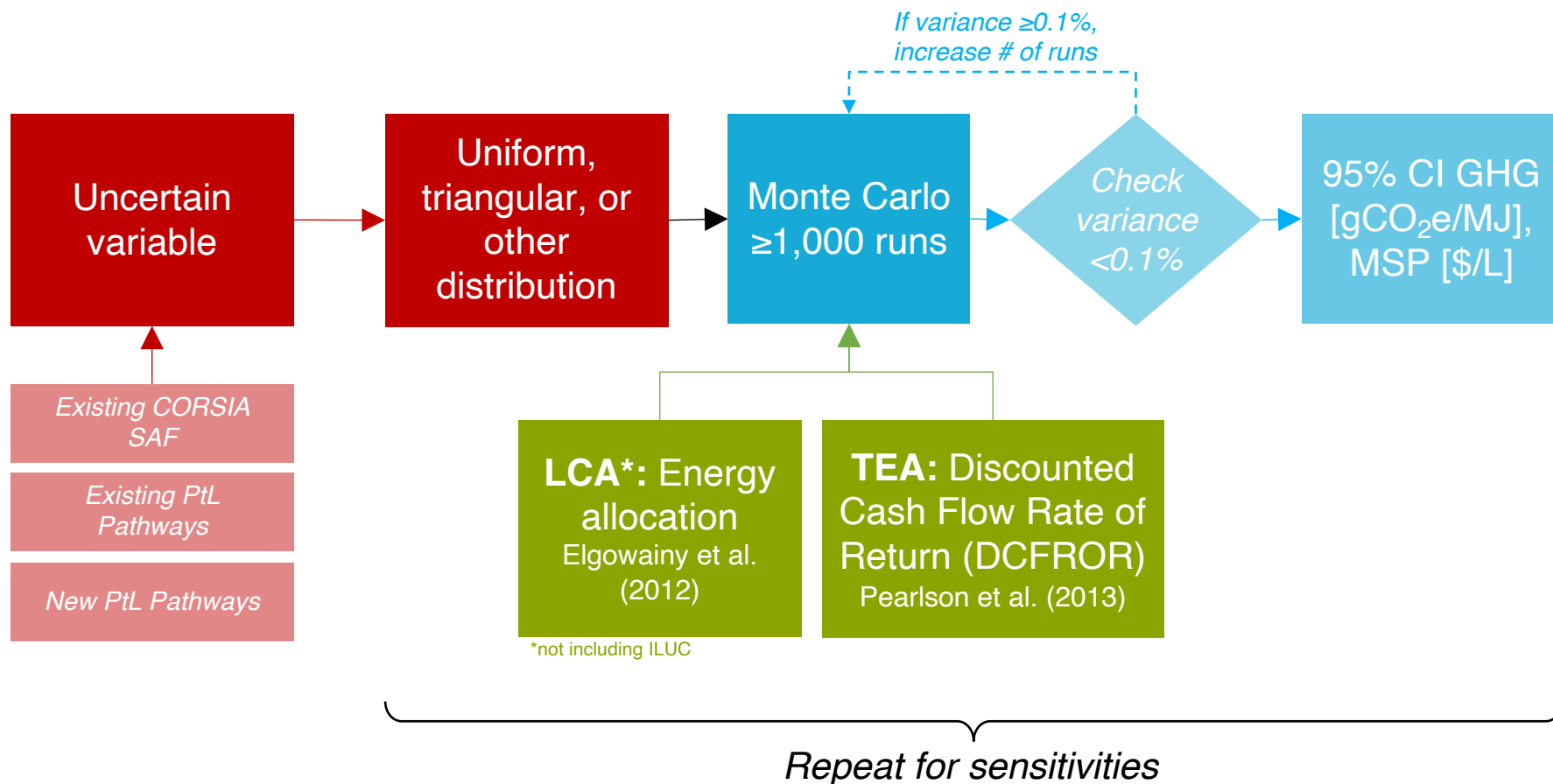
To be captured as needed (e.g. for discussing different electricity inputs)

Sensitivity

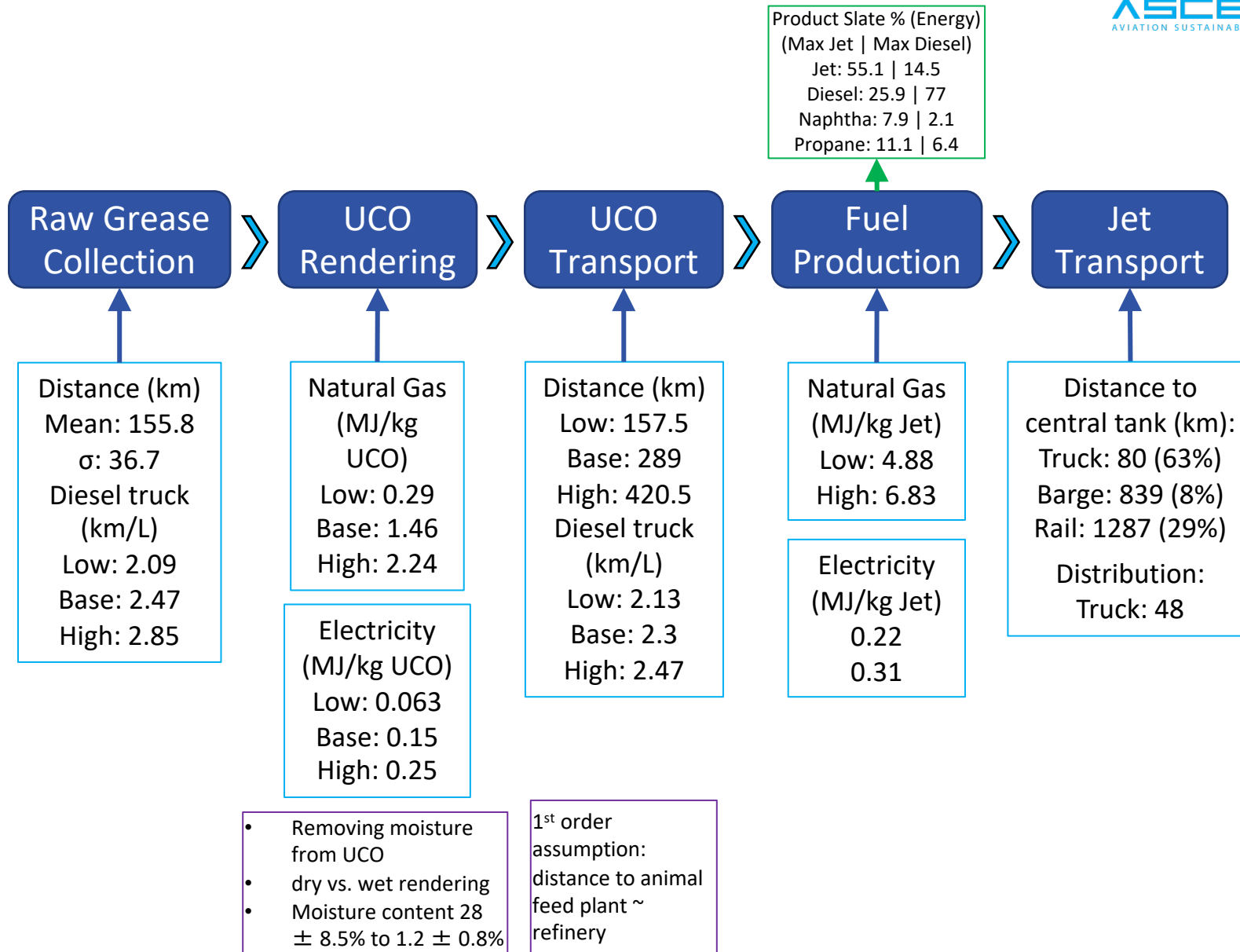
Studying impact on dependent variable due to variability

Example

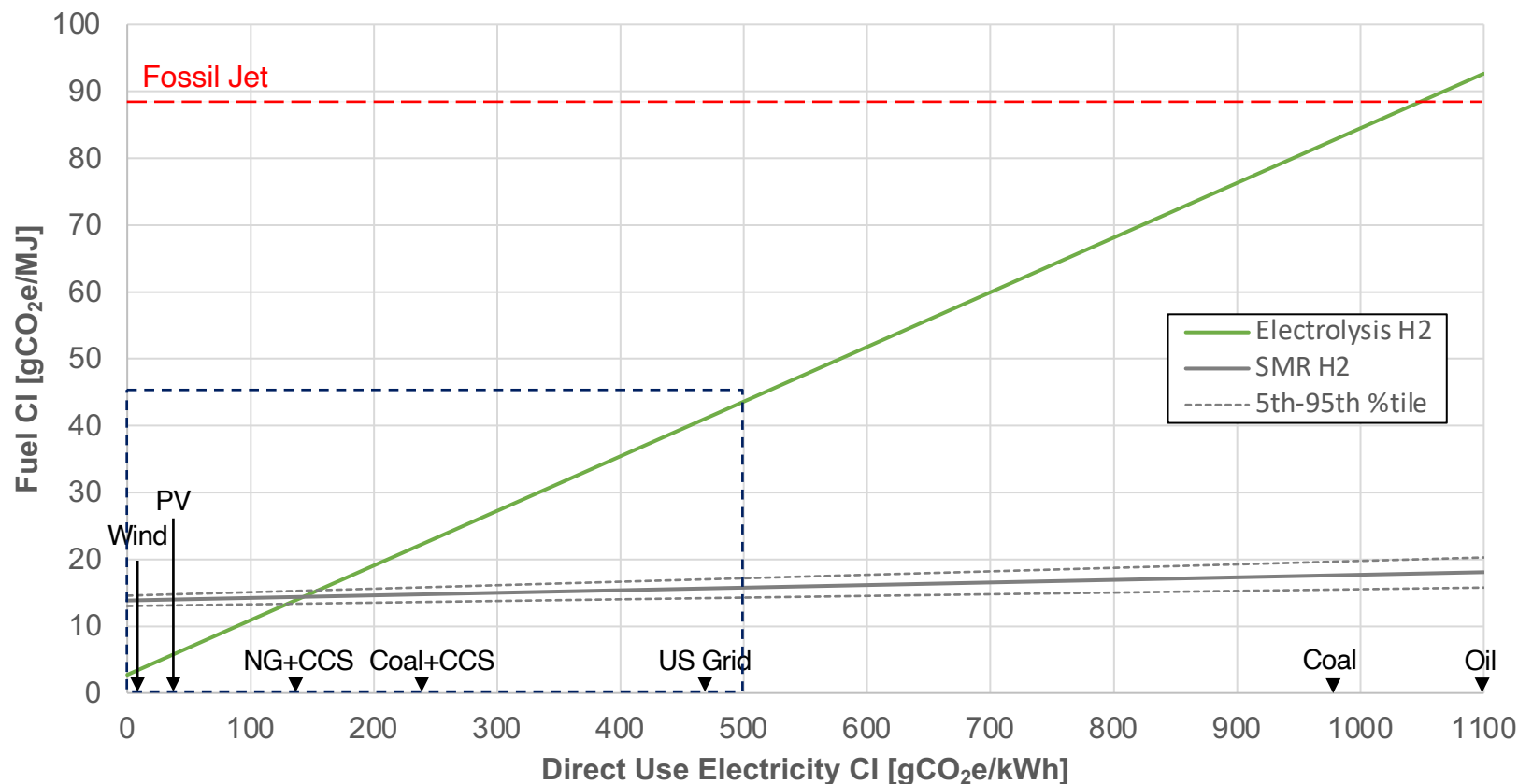
Impact on GHG emissions from differences in electricity carbon intensity



Example: HEFA UCO stochastic inputs



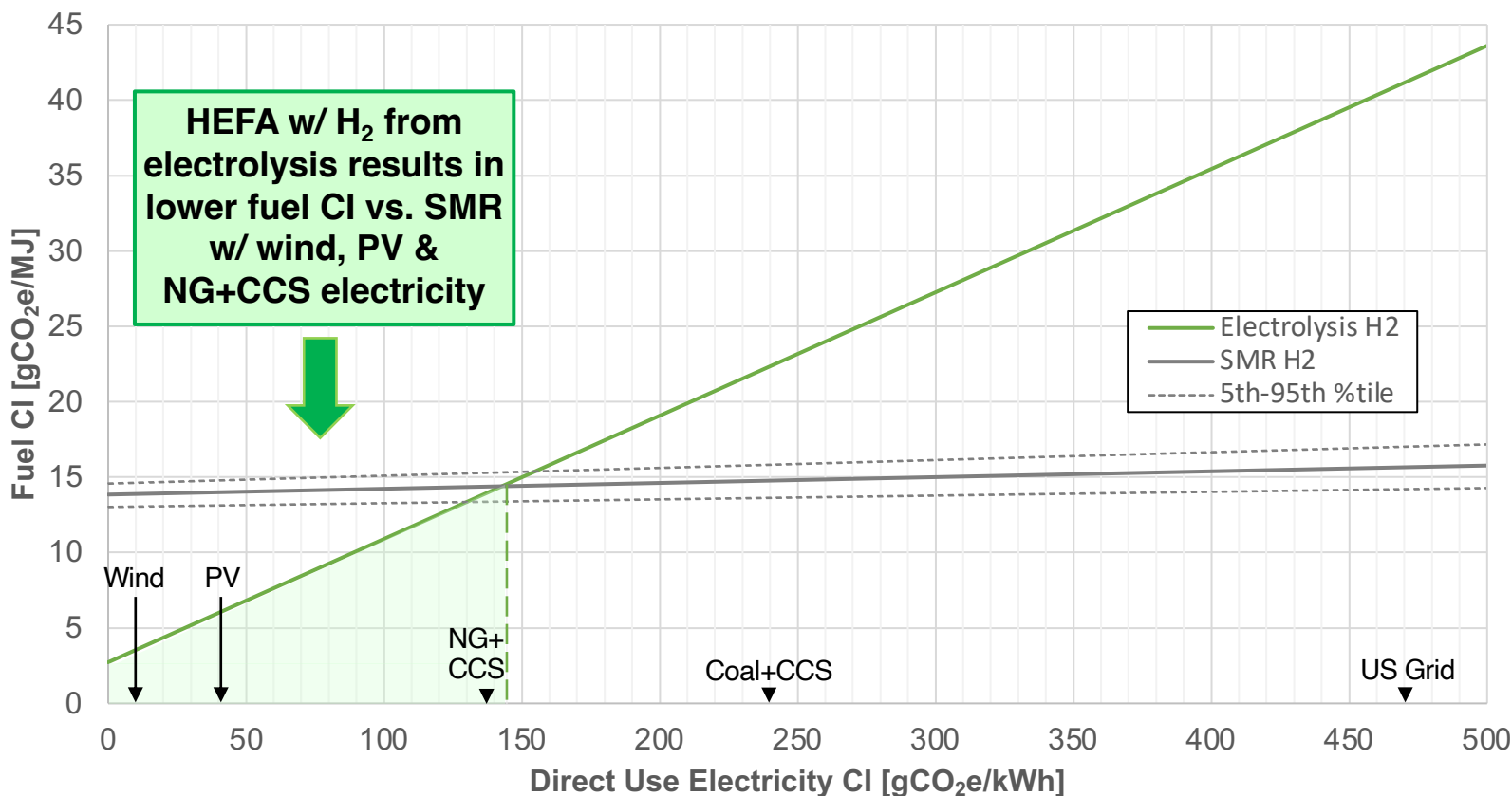
HEFA UCO Fuel CI vs. Electricity CI Sensitivity



- ~5% of direct process energy is electricity for HEFA UCO, thus sensitivity to electricity CI is low (see *grey SMR H₂ line*)
- When adding hydrogen from renewable electricity, the CI sensitivity increases significantly (see *green Electrolysis H₂ line*)

HEFA UCO stochastic LCA: H₂ source (SMR & Electrolysis) and sensitivity to electricity carbon intensity

HEFA UCO Fuel CI vs. Electricity CI Sensitivity



- HEFA UCO w/ H₂ from electrolysis results in lower fuel CI vs. SMR w/ electricity <143gCO₂e/kWh
- Other possible sources: Hydro, Geothermal, Willow IGCC

Conclusions

- We have presented the SAF production problem to visualize the environmental, economic, and technical tradeoffs and to guide us in technology development
- In the case of syngas it imposes constraints into the design of systems that maximize carbon conversion efficiency
- Stochastic analysis for novel fuel pathways is important for fuel LCA/TEA to establish uncertainty
- Sensitivity studies, especially with electricity, will be useful for current and future PtL pathways to quantify the range of possible emissions reductions

Next Steps

- Develop a tool to estimate production costs and expected incentives if the yield of products and process inputs are known
- Propose new SAF production technologies integrated with different hydrogen production concepts
- Continue LCA/TEA model development and apply to more existing CORSIA-approved SAF pathways & feedstocks
- Adjust model for newly identified PtL pathways