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Developing Thermal Control of *Salmonella* in Low-Moisture Foods Using Predictive Models

Salmonella is highly resistant to heat inactivation in low-moisture food systems, but pilot-scale thermal treatments show promise for combatting it

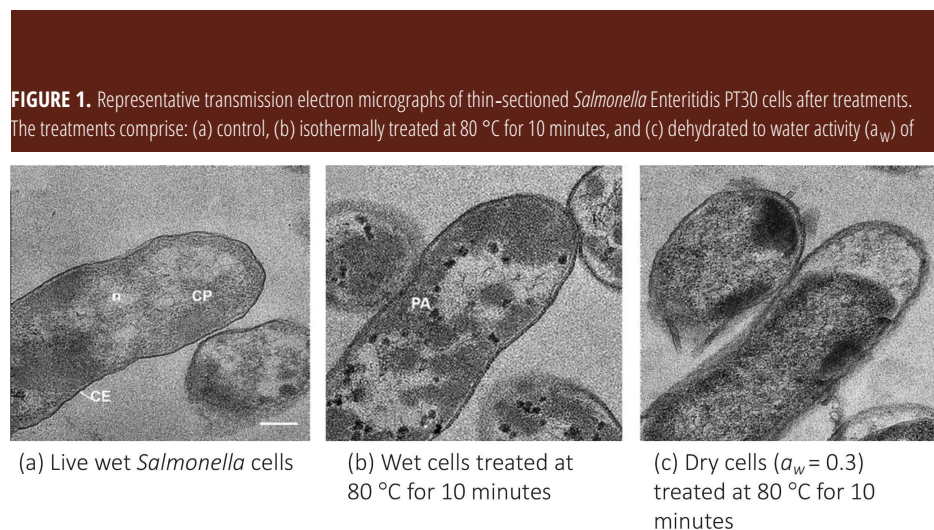
By Ren Yang, Ph.D., Postdoctoral Research Associate, and Juming Tang, Ph.D., Regents Professor, Department of Biological Systems Engineering, Washington State University

Over the past decades, concerns over the safety of low-moisture foods have grown due to numerous outbreaks and recalls associated with various products. Among the involved pathogens, *Salmonella* spp. emerged as a prevalent concern.¹ While thermal pasteurization has effectively reduced the risk of pathogen contamination in high-moisture foods, vegetative bacterial pathogens like *Salmonella* spp. have shown high thermal tolerance in low-moisture foods. The thermal processing conditions designed for high- and intermediate-moisture foods are inadequate for achieving sufficient inactivation of bacterial pathogens in low-moisture foods. Therefore, it is crucial to gain a comprehensive

This article presents the latest research findings that shed some light on the reasons behind the enhanced thermal resistance of foodborne pathogens in low-moisture food systems. It discusses the quantitative relationships between thermal treatment temperature, water activity, and relative humidity and their impact on bacterial pathogens' log-reduction rate (*D*-value). Furthermore, it presents case studies that demonstrate the development and validation of predictable thermal treatments for *Salmonella* control in pre-packaged food materials, like wheat flour, and on the surfaces of food particles, such as peppercorns.

Thermal Tolerance of Bacterial Cells in Low-Moisture Environments

The thermal death of bacteria occurs through irreversible changes in critical cell components.² Desiccated vegetative bacteria cells have demonstrated remarkably high thermal resistance, even comparable to that of spore-formers like *Bacillus* and *Clostridium*.³ While the role of water in the thermal death of microorganisms is not yet clearly understood, the absence of water likely restricts molecular mobility and protein deformation in living cells, thereby reducing the rate of denaturation of the cell's ultrastructure and prolonging the time required to cause irreparable damage.^{2,4,5} [Figure 1](#) illustrates how desiccated *Salmonella* cells did not exhibit clear protein aggregations as did wet bacterial cells after undergoing the same thermal treatment (80 °C, 10 minutes).



Correlating Thermal Resistance Data with Measurable Parameters

The thermal death kinetics of bacteria are commonly modeled using a first-order reaction, which suggests that each bacterial cell has the same probability of dying under a specific treatment condition.⁶ Temperature was a key factor influencing the reaction

rate, and its impact on the rate coefficient, in general, follows the Arrhenius equation.^{7,8} Early studies focused on sterilization or pasteurization of high- or intermediate-moisture foods and did not include the influence of food water activity. Research on pathogen control in low-moisture foods has significantly increased over the past two decades. It is now widely recognized that the thermal death of bacteria, such as *Salmonella*, *Listeria*, and *E. coli*, in low-moisture foods can also be effectively modeled using first-order kinetics.⁹ At any given temperature, however, the thermal inactivation rates for the bacteria in low-moisture foods were much lower than those in high moisture-content foods.¹⁰

Efforts have been made to establish connections between bacteria's thermal death time (*D*-value) and measurable environmental factors, including temperature and water activity (a_w), to guide the design of thermal processes for low-moisture food commodities. In early studies, water activity was considered a temperature-independent factor when measuring the thermal resistance of bacteria. While this assumption may be true for high- or intermediate-moisture food matrices that contain free water, it does not apply to low-moisture foods that have multilayers of water molecules bonded with different levels of energy.¹¹

Figure 2 summarizes recently published data regarding changes in the water activity of different materials with temperature. The water activity of almond flour and wheat flour increases with temperature at different rates, while the water activity of peanut oil decreases with temperature. Figure 2 also includes the change in the water activity of the biomass of *Salmonella* cells. *Salmonella* demonstrates its own water activity trend that differs from the three food materials shown in the same figure. That is, bacterial cells gained moisture in wheat flour and lost moisture in almond flour and peanut oil when heated from room temperature to 80°C. As illustrated in Figure 3, the vapor pressure difference between bacterial cells and their surrounding environments will change the moisture content of the bacterial cells to reach an equilibrium with that of the environmental water activity. Such a process is estimated to take less than seconds.¹² By measuring the high-temperature water activity of low-moisture foods and the *D*-value of *Salmonella* in a food matrix, it is possible to establish a quantitative relationship to predict the required pasteurization time at a specific temperature.¹⁰

FIGURE 2. Water activity changes differently in *Salmonella* cells, wheat flour, almond flour, and peanut oil with temperature increase, causing humidity to rebalance during thermal processing.

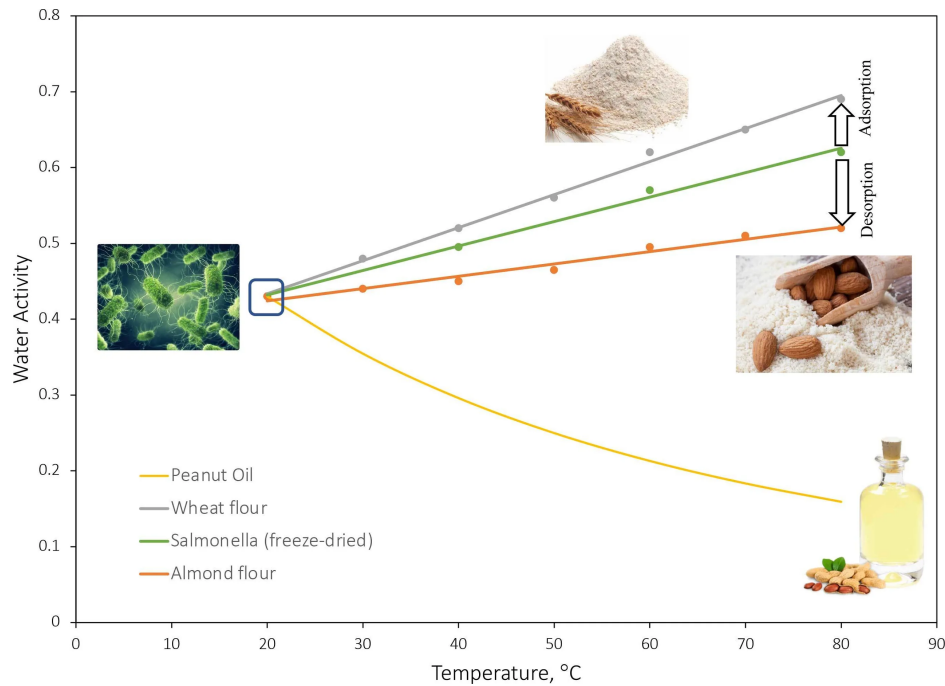
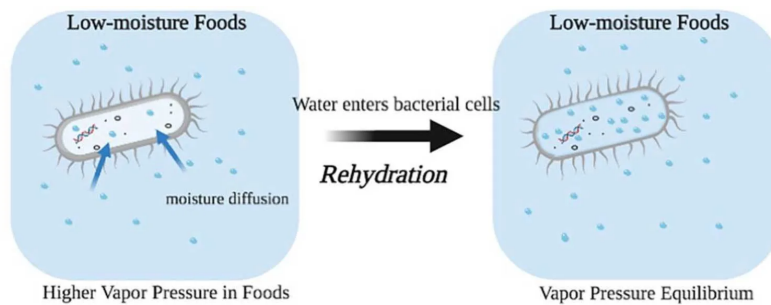
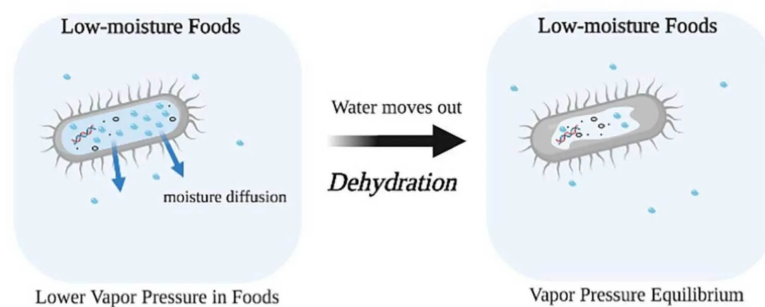


FIGURE 3. Illustration of moisture exchange between *Salmonella* cells and different microenvironments during thermal processing.¹³



(A) Higher a_w in low-moisture foods than bacterial cells at thermal-treatment temperature

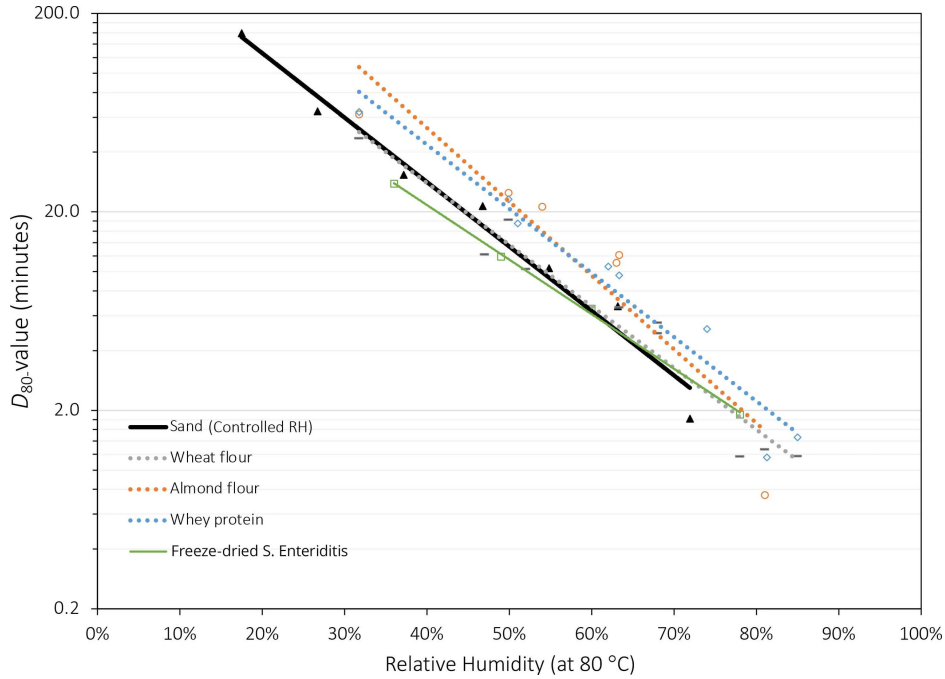


(B) Lower a_w in low-moisture foods than bacterial cells at thermal-treatment temperature

Several published studies reported quantitative relationships between D-values of *Salmonella* and high-temperature water activity of different low-moisture food matrices, including chocolate,¹⁴ peanut butter,¹⁵ dehydrated spices,¹⁶ egg powders,¹⁷ wheat flour, almond flour, and whey proteins.¹⁸ Figure 4 illustrates how the high-temperature water activity of the food matrix or the relative humidity during thermal treatment plays a significant role in determining the thermal resistance of *Salmonella* in low-moisture foods. These data can be utilized to predict the D-value of *Salmonella* in different low foods. For example, in the cases presented in Figure 2, when wheat flour with water activity of 0.43 is heated in sealed containers from 20 °C to 80 °C, the water

activity of the material will change to 0.69. The D -value of *Salmonella* in the wheat flour at 80 °C will be 4.9 minutes. For almond flour, however, the water activity will change from 0.43 at 20 °C to 0.52 at 80 °C, and the corresponding D -value of *Salmonella* at 80 °C will be 21.2 minutes. For peanut oil, the water activity will change from 0.43 at 20 °C to 0.12 at 80 °C, and the D -value of *Salmonella* at 80 °C will be above 160 minutes.

FIGURE 4. Thermal resistance of *Salmonella* Enteritidis PT30 at 80 °C in LMF corresponding to high-temperature water activity (transferred to relative humidity, or RH) and in sand under controlled RH.^{13,18,19} The equation was generated from linear regression using the data of sand.



A mathematical equation describing the thermal resistance of a specific pathogen in low-moisture foods (adapted from Gaillard *et al.*²⁰) can be used to help the design of effective thermal processing operations ([Equation 1](#)):

EQUATION 1.

$$\log D = \log(D_{ref}) - \frac{T - T_{ref}}{Z_T} - \frac{RH - RH_{ref}}{Z_{RH}} + MF$$

where D_{ref} is the D value at the reference temperature, T_{ref} , and the reference humidity, RH_{ref} ; Z_T and Z_{RH} indicate the temperature and RH increase that leads to a tenfold reduction in D -value, respectively; MF is the matrix factor showing the effect of food composition.

The reported Z_T and Z_{RH} values of *Salmonella* spp. are summarized in [Table 1](#) and [Table 2](#). The data reveals that Z_T ranged between 6.7 °C and 20.6 °C, indicating a clear dependence on relative humidity where higher RH values correspond to lower Z_T values. The lowest Z_T value was observed in whey protein at 84–86 percent RH and 60–75 °C, measuring only 6.7 °C, which is merely 1–2 °C higher than the value reported for *Salmonella* in high-moisture meat products.²¹ Generally, Z_T values for *Salmonella* spp. in low-

moisture foods exceed 10 °C. These values are comparable to the range of thermophilic spores in high-moisture foods (7–16 °C).²² This suggests that the thermal resistance of vegetative bacterial cells in low-moisture foods exhibits a similar sensitivity to temperature changes as that of spores in high-moisture foods.

Z_{RH} indicates the increase in relative humidity required to cause a tenfold reduction in the D -value of *Salmonella*. Currently reported Z_{RH} for *Salmonella* in low-moisture foods ranges between 21 percent and 41 percent at 80 °C. These values are higher than the Z_{RH} value of *Bacillus* spores at 85–105 °C and 80–100 percent RH, which is 16 percent. It has been reported that Z_{RH} may decrease significantly as RH surpasses 70–80 percent;²³ however, further investigation is needed to validate this observation and provide guidance regarding the thermal death of bacteria at higher humidity levels.

TABLE 1. Z_T values of *Salmonella* spp. in low-moisture foods at specific ranges of application

Food Matrix	Range of Application		Z_T , °C	Reference
	Temperature, °C	RH, percent		
Milk chocolate	70–80	33	18.8 ± 2.5	Sun <i>et al.</i> ¹⁴
	70–80	44	20.6 ± 4.1	
	70–80	52	18.1 ± 0.5	
Peanut butter	70–100	33	15.4	Yang, Wei, <i>et al.</i> ¹⁵
	70–100	53	12.6	
Soy protein powder	80–99	25–32	12.5	Jin <i>et al.</i> ²³
	80–99	36–43	13.2	
	80–99	47–52	13.1	
	75–95	52–58	11.6	
	70–90	58–62	11.2	
	70–85	68–70	10.8	
	70–85	74–76	8.0	
	60–75	84–86	6.7	

TABLE 2. Z_{RH} values of *Salmonella* spp. in low-moisture foods at specific ranges of application

Food Matrix	Range of Application		Z_{RH} , percent	Reference
	Temperature, °C	RH, percent		
Black peppercorn	80	60–80	21.3	Yang <i>et al.</i> ²⁷
Soy protein powder	70	58–74	44	Jin <i>et al.</i> ²³
	75	52–75	39	
	80	25–69	41	
Wheat flour	80	31–78	32.2	Xu <i>et al.</i> ¹⁸
Almond flour	80	32–81	28.9	Xu <i>et al.</i> ¹⁸
Whey protein	80	32–85	30.9	Xu <i>et al.</i> ¹⁸
Sand (SiO ₂)	80	18–72	31	Liu, Tang, <i>et al.</i> ^{19,26}

Case Studies: Applying Valid Thermal Processing for Low-Moisture Foods Pathogen Control

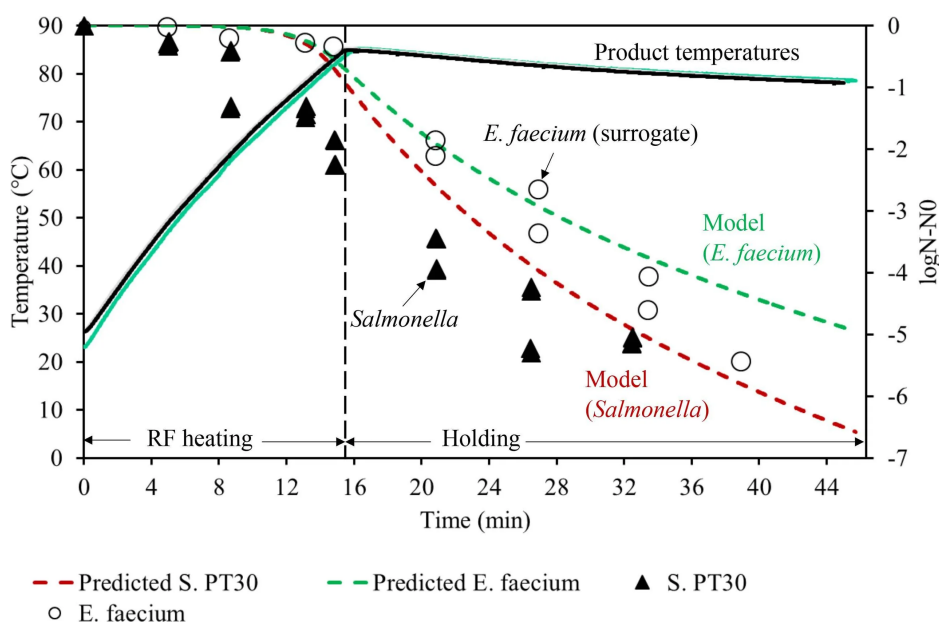
Thermal processing of low-moisture foods involves intricate heat and mass transfer interactions among the food matrix, heat-conducting medium, microorganisms, and the surrounding gaseous phase. The actual temperature and humidity conditions experienced by microorganisms during thermal processing can vary due to factors such as heating method, equipment design, material properties (e.g., moisture content), and operational practices. To illustrate the practical implementation of our developed method for predicting microbial thermal resistance in low-moisture foods

RF Heating for In-Package Pasteurization of Wheat Flour

To pasteurize wheat flour (e.g., in 5-lb bags) with different moisture contents (e.g., 11.6 percent and 14.5 percent) using an RF heating process (volumetric heating), the holding time required at a specific target temperature can be predicted by measuring or calculating the high-temperature water activities of the flour samples, and then determining their corresponding D -values for *Salmonella*. In this case, calculations indicate that the water activities at room temperature for the two batches of flour are 0.43 and 0.64, which would increase to 0.69 and 0.82, respectively, at 80 °C.^{24,25} According to the equation shown in [Figure 4](#), the D_{80} values of *Salmonella* at these water activity levels are 3.2 and 1.2 minutes, respectively. To achieve a 5-log reduction, the two flour batches must be held at 80 °C for 16 and 6 minutes, respectively.

Alternatively, a higher temperature can be employed to reduce the holding time. For example, assuming a Z_T value of 11 °C, the holding time can be reduced to 0.32 and 0.12 minutes if the sample is heated to 91 °C. [Figure 5](#) illustrates an example of wheat flour pasteurization using RF heating and holding. The flour in the package (3 kg, $a_w = 0.45$) was heated to 83 °C in 15 minutes and then held for a specific duration to achieve the desired level of lethality. As shown in [Figure 5](#), the lethality contributed by the heating section can also be counted using a model to calculate the accumulated lethality ([Equation 2](#)).

FIGURE 5. Survival and model prediction of *Salmonella* and *E. faecium* in wheat flour in RF heating and holding sections of a pasteurization process.²⁶



where F is the total equivalent heating time at the reference temperature; T is the real-time temperature at time t , and D_{ref} is the D -value at the reference temperature, T_{ref} . The log reduction in a process can be calculated as shown in [Equation 3](#):

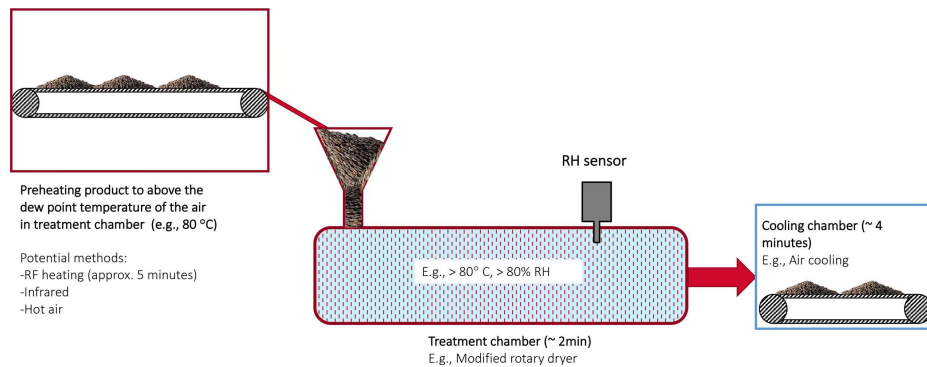
EQUATION 3.

$$\log \frac{N_0}{N} = \frac{F}{D_{T_{\text{ref}}}}$$

Controlled Humidity Thermal Pasteurization of Dehydrated Food Commodities

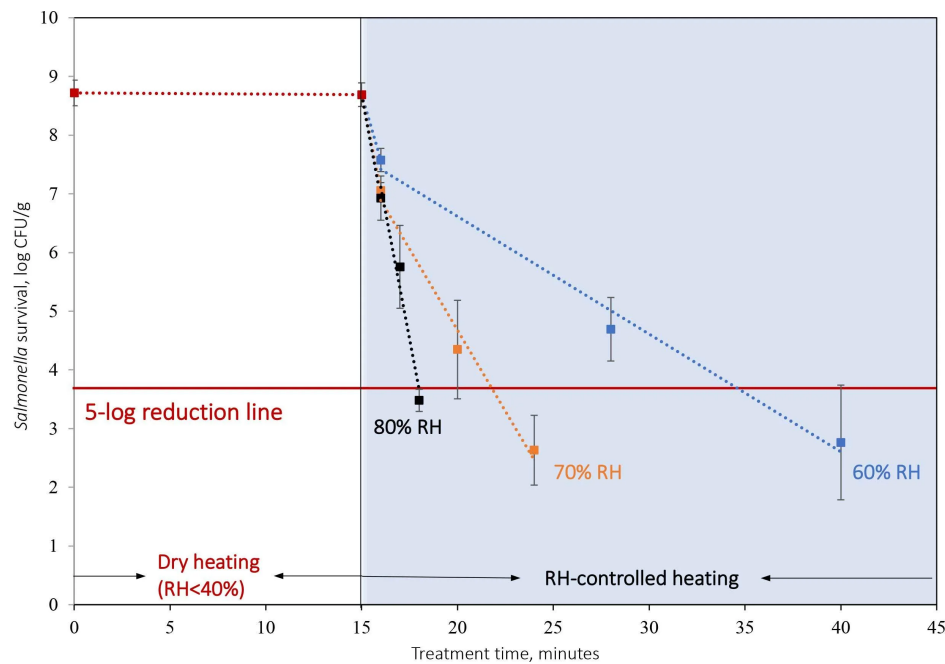
Our recent research has demonstrated the effectiveness of a highly efficient pasteurization process for low-moisture foods utilizing controlled humidity.²⁷ In this process, low-moisture food commodities were preheated using conventional heating methods and subsequently transferred to a vessel with controlled temperature and humidity for a designated holding time ([Figure 6](#)). The purpose of preheating is to raise the product temperature above the dew point of the air in the treatment chamber, preventing condensation. After achieving the target level of lethality in the treatment chamber, the product was cooled using ambient air.

FIGURE 6. Concept of Controlled-Humidity Thermal Pasteurization Process



The authors' lab-scale unit has exhibited significant potential for pasteurizing various low-moisture food commodities ([Figure 7](#)). A 15-minute preheating (RH < 30 percent) resulted in negligible reduction of *Salmonella*, whereas the introduction of humid air (80 °C, RH between 60 percent and 80 percent) led to rapid bacterial reduction (a 6-log reduction in just 3 minutes at 80 percent RH). The reduction follows first-order kinetics, depending on temperature and humidity levels. This study provides a method for treating different low-moisture food commodities without the need to consider their initial moisture conditions. As discussed previously, humidity control plays a critical role in regulating the moisture content of bacteria on food particle surfaces. Preheating ensures that no condensation occurs when introducing humid hot air, potentially eliminating the need for post-treatment drying. By

FIGURE 7. Survival populations of *Salmonella* Enteritidis PT 30 in black peppercorns at different thermal treatment steps combining 15-minute dry heating and RH-controlled heating at designated durations ($n = 3$).²⁷



Conclusion

Enhancing thermal processing methods for pathogen control in low-moisture foods is crucial for ensuring public health and the safety of food products. The thermal resistance of bacterial pathogens, particularly *Salmonella*, is influenced by various factors, including temperature, water activity, and relative humidity. Understanding these factors and their quantitative relationships enables the development of effective thermal treatments that can mitigate the risk of pathogen contamination in low-moisture food systems. The case studies in this article highlight the importance of considering high-temperature water activity and humidity control in designing valid thermal processes for low-moisture food pathogen control. By employing these scientific findings, the food industry can implement robust strategies to minimize the risk of outbreaks and recalls associated with low-moisture food commodities. ■

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